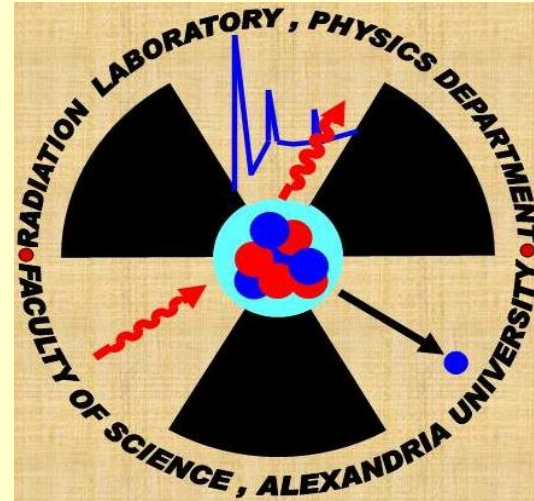


Phy 310
Detectors
and
Accelerators

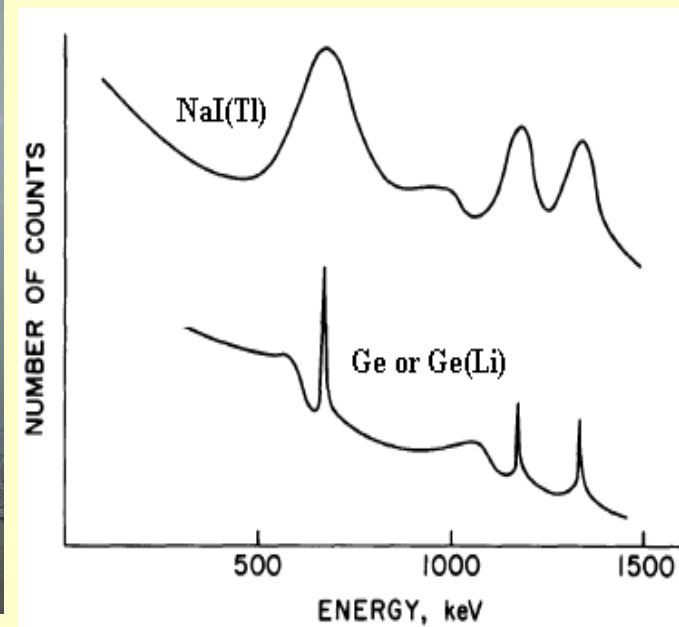
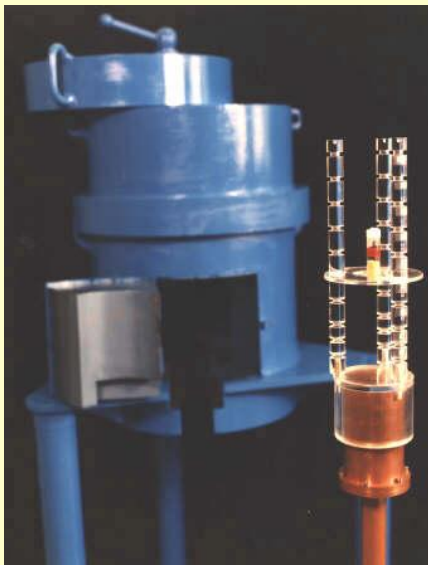


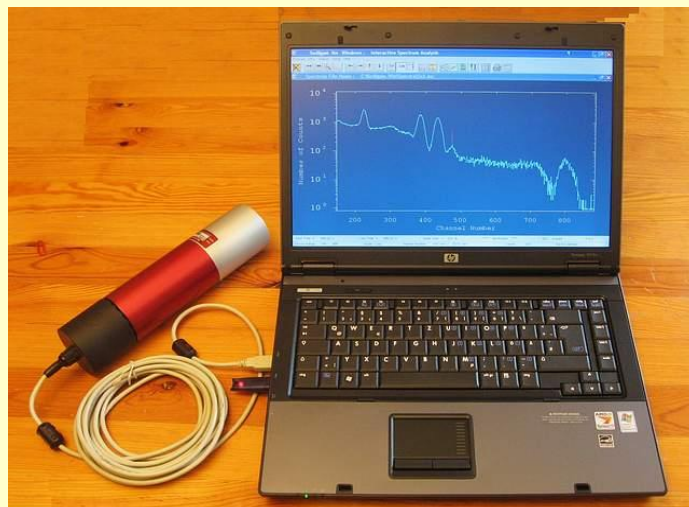
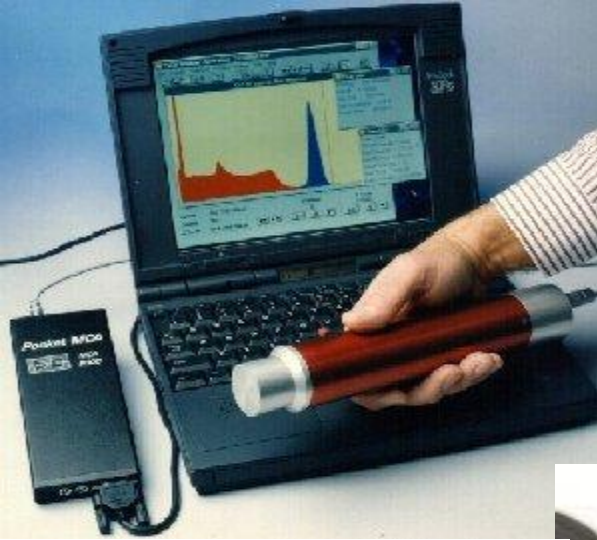
Dr. Mohamed Salem

1- Detectors: General Properties of Radiation Detectors. Ionization Chambers. Proportional counters. Geiger Muller Counters. Scintillation Detectors , Radiation Spectroscopy with Scintillation. Semiconductor Diode Detector. Germanium Gamma Ray Detectors, Lithium-Drifted Silicon Detectors. Slow Neutron Detection Methods. Miscellaneous Detector Types.

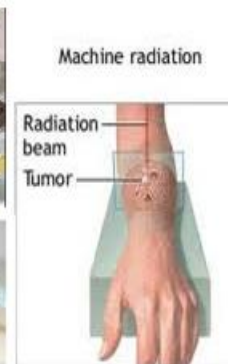
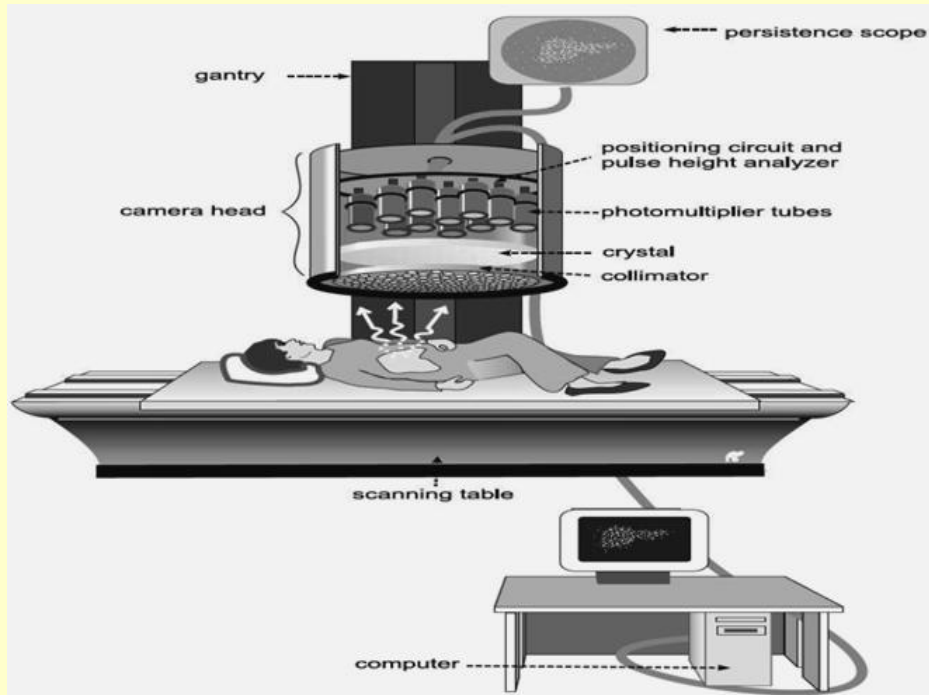
2- Accelerators: Ion Source. Electrostatic Accelerators. Cockcroft-Walton Accelerator, Van de Graaff accelerator, Tandem Van de Graaff Accelerator. Magnetic Resonance Accelerator. Cyclotron Accelerators, Synchrocyclotron Accelerators and Synchrotron Accelerators. Linear Accelerators. Colliding Beam Accelerators.

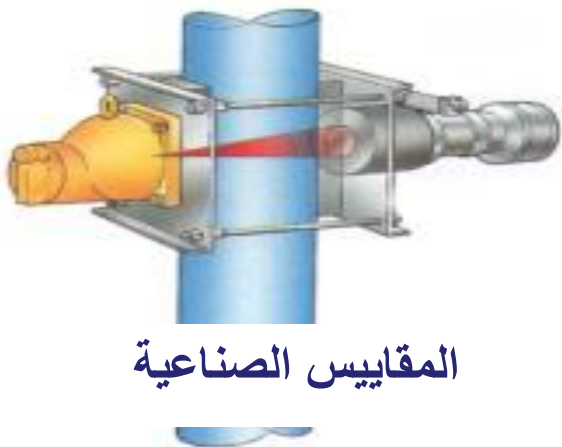
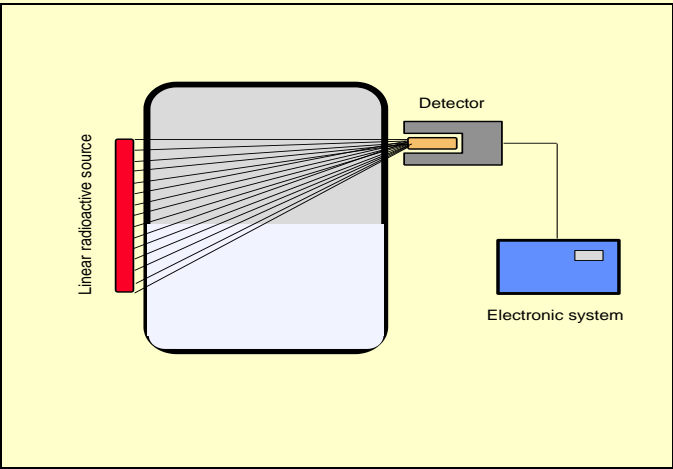
Some Applications





Gamma Camera





المقاييس الصناعية

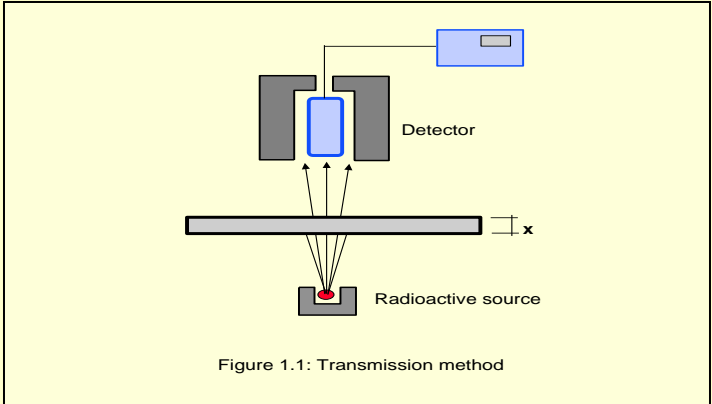
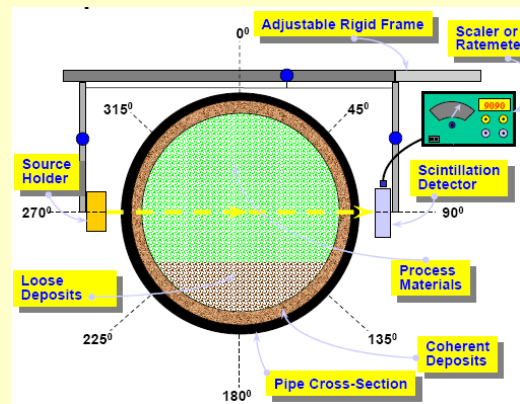
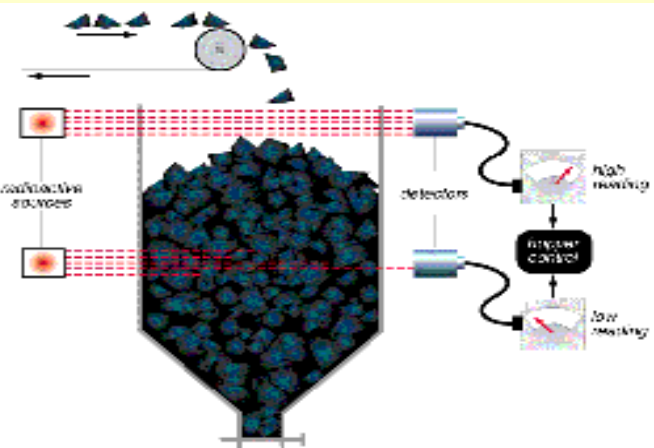
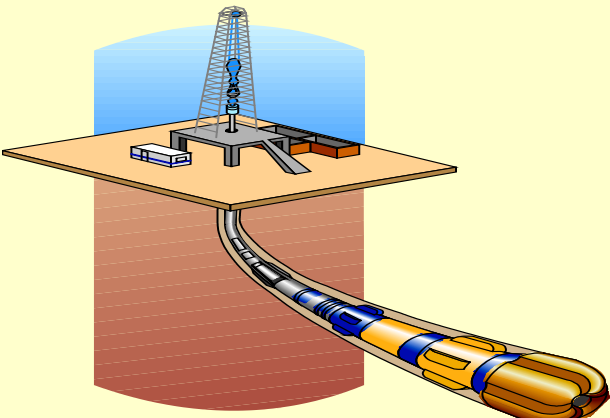
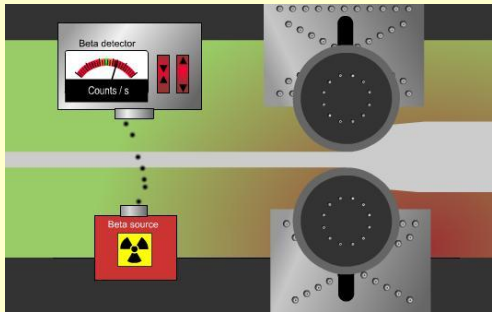
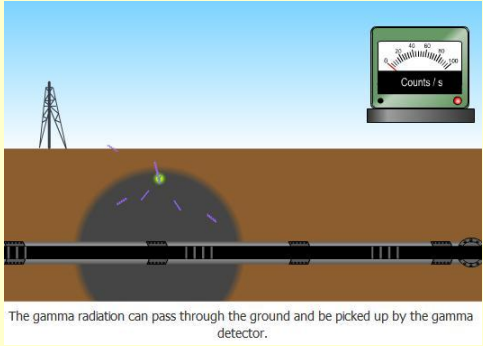
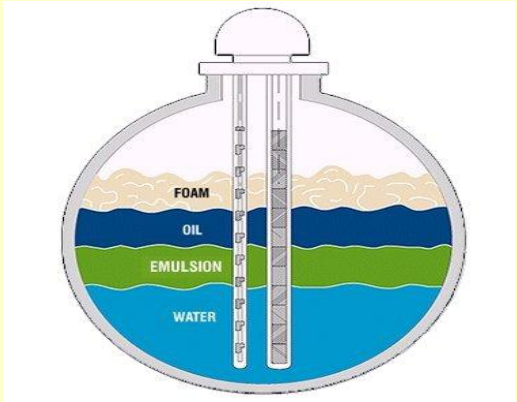
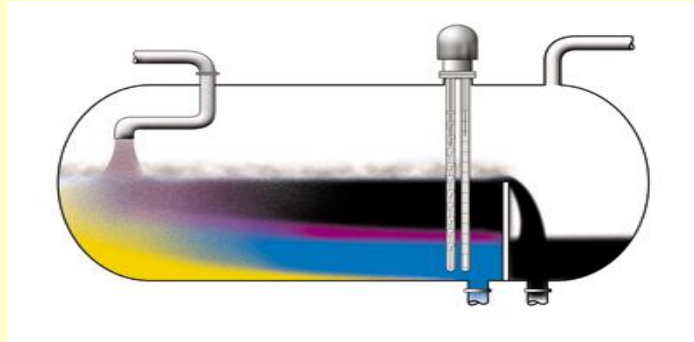


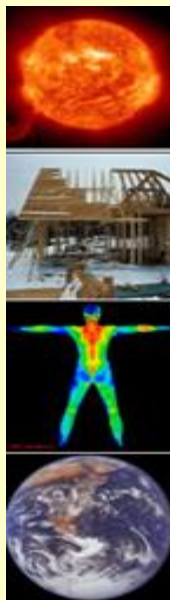
Figure 1.1: Transmission method

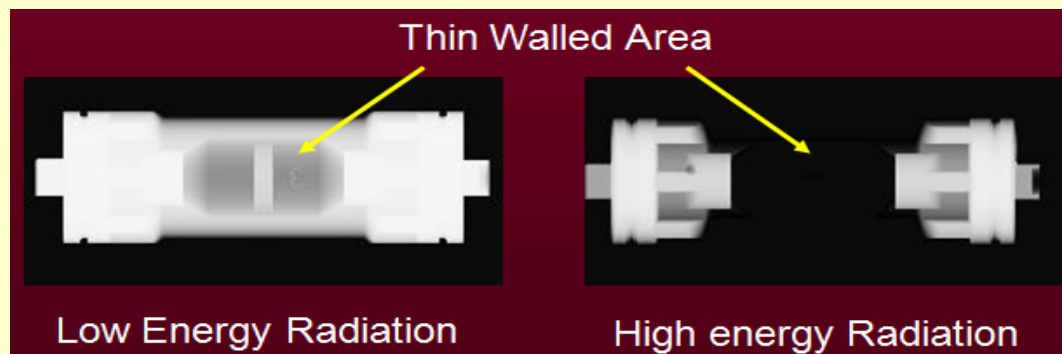
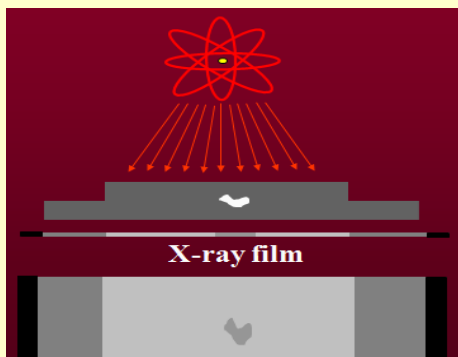
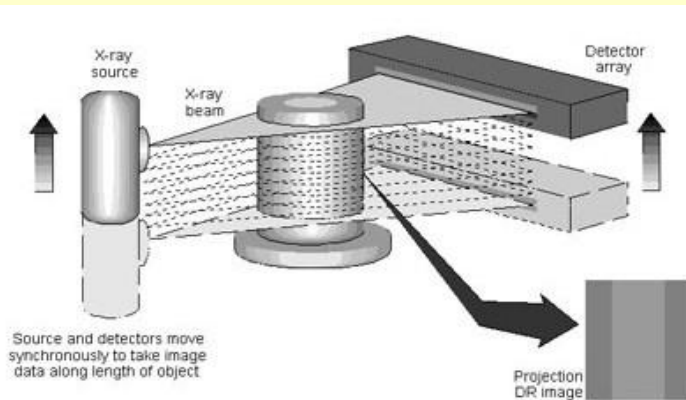
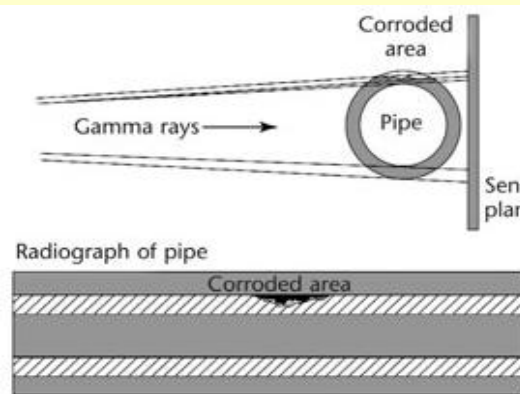
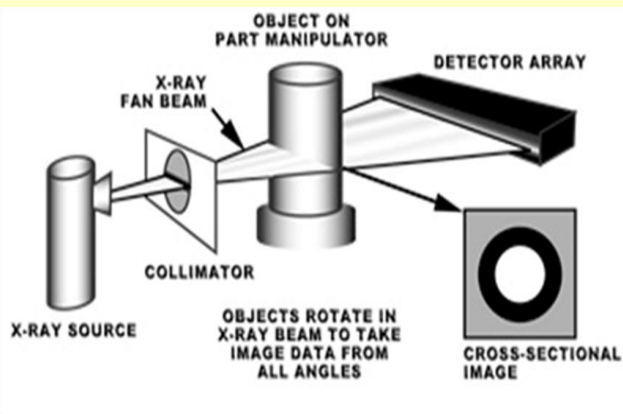
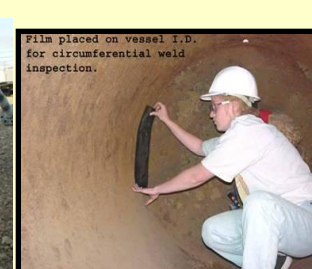


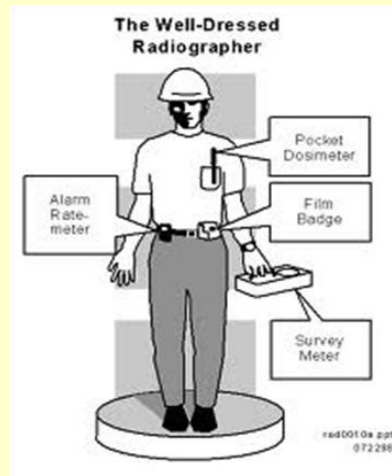
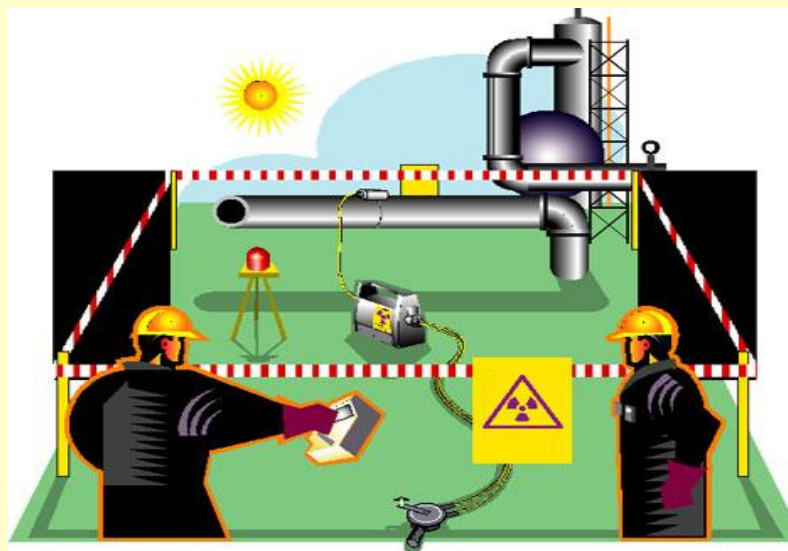


الجرعات الإشعاعية التي يتعرض لها الإنسان

أ - مصادر إشعاعية طبيعية		الجرعة الإشعاعية المكافئة مللي سيفرت/سنة
1	الأشعة الكونية والخلفية الإشعاعية عن سطح البحر	0.46
ب - مصادر إشعاعية صناعية		
1	مشاهدة التلفاز الملون لمدة 3 ساعات	0.06
2	عدد 2 صورة أشعة سينية للصدر	0.16
3	عدد 2 صورة أشعة سينية للمعدة	4.88
4	عدد 2 صورة أشعة سينية للأسنان	0.08
5	عدد 2 صورة أشعة سينية للجمجمة	0.44







Survey Meter



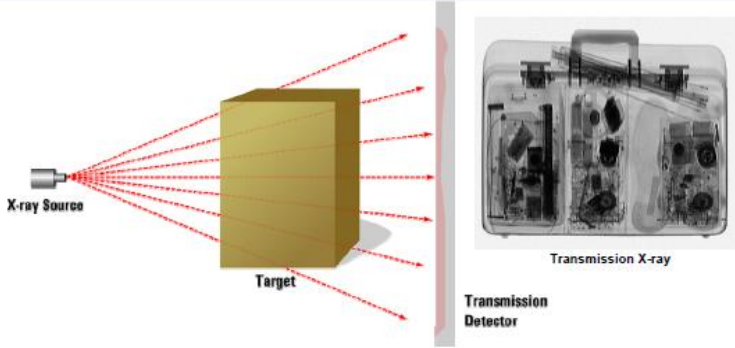
Pocket Dosimeter



Radiation Alarm



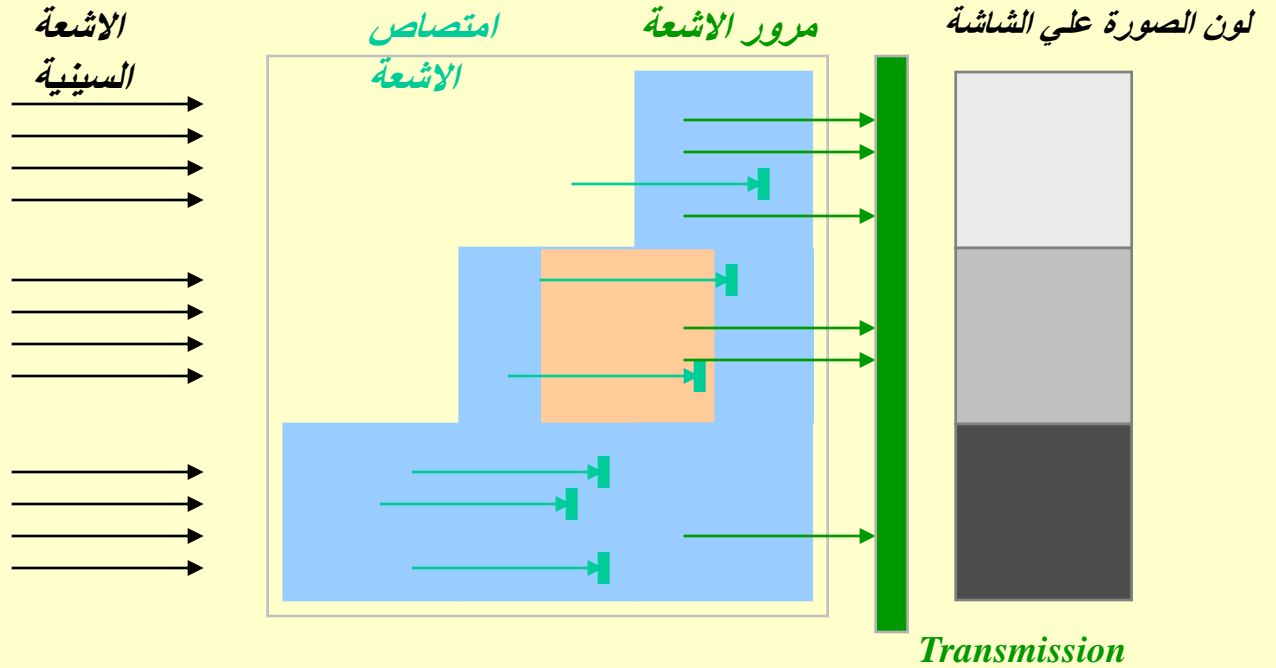
Radiation Badge



Transmission X-rays detect by passing an X-ray beam through a target to a detector on the far side.

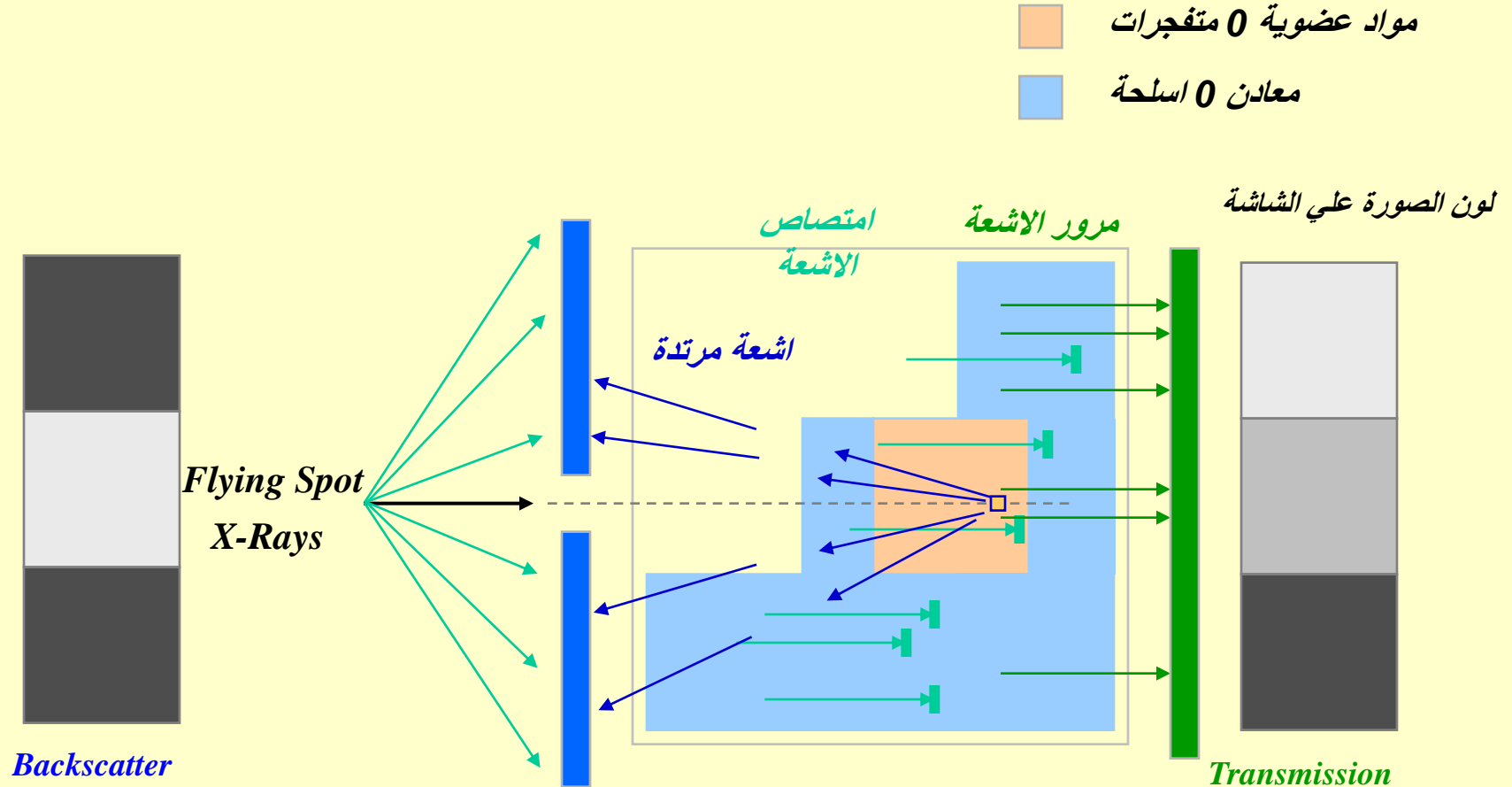
Transmission Imaging

- المواد العضوية المتفجرات
- المواد المعدنية الاسلحة



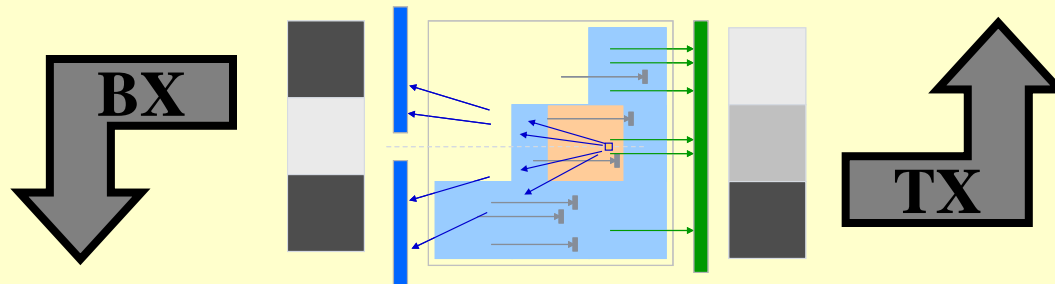
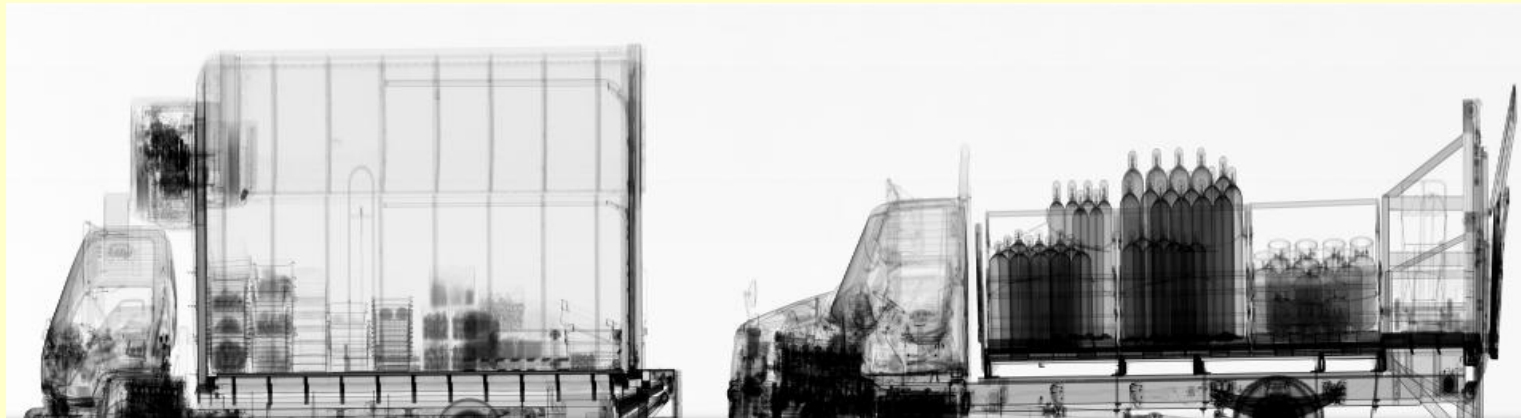
صور الكثافات المختلفة للأشعة النافذة

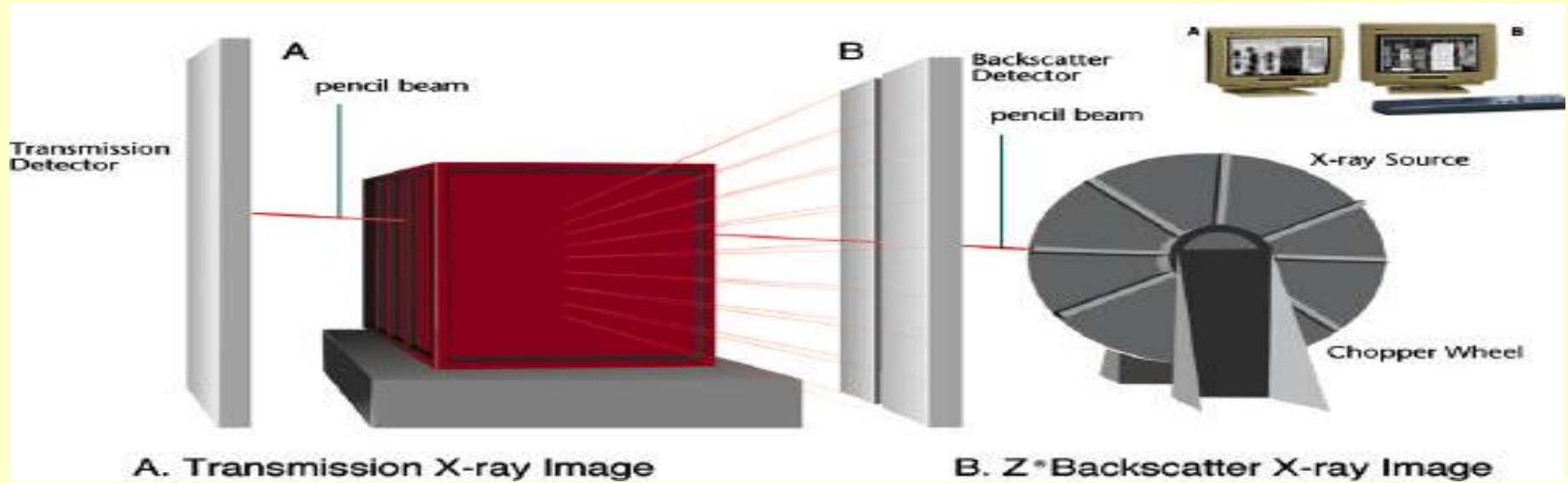
Backscatter Imaging



صور للمواد المختلفة للأشعة المرتدة

Transmission and Backscatter Images





الطاقة

من 1 الي 9 مليون الكترون فولت



الطاقة

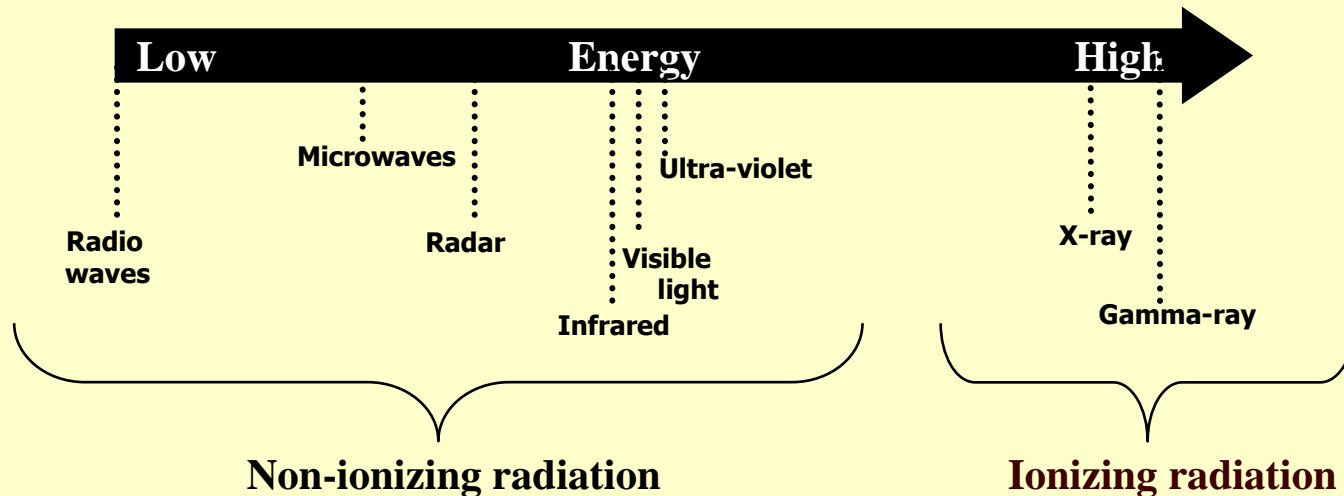
من 50 الي 500 كيلو الكترون فولت



x-rays and gamma rays

Ionization radiation

Electromagnetic radiation



☐ Individual wave packets of this radiation are called **photon** ☐

■ **The main difference between gamma and x-ray is their origin** ■

While **x-ray** are produced by **atomic excitation** or from an **electron** as it changes direction when passing an atomic nucleus; this latter type of x-ray is called **bermsstrahlung radiation**, **gamma rays** are emitted by **transition from excited states in the nucleus**.

Applications of gamma rays spectroscopy

In radiation physics, **measuring and studying the energy gamma spectra emitted from radioisotopes** are very important and have many applications.

- 1) **Identification of the radioactive isotopes.**
- 2) **Study of the nuclear structure.**
- 3) **Measuring the absorbed doses**
- 4) **Determination of the interaction cross section.**

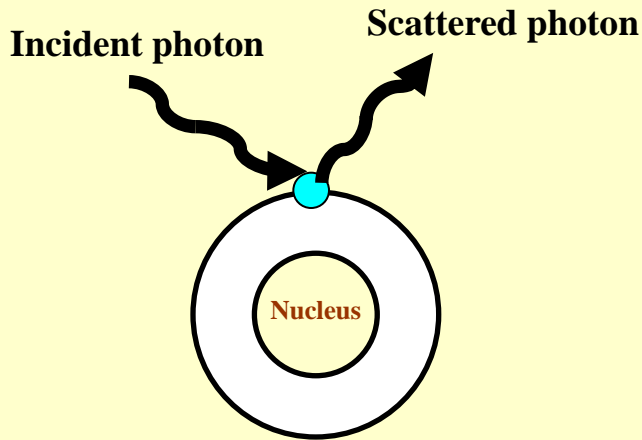
■ To get a spectrum with high accuracy we must have ■

- 1) **Detection and recording system “method of measurements”.**
- 2) **Good values for detectors factors that concern the measurements analysis “efficiencies”.**

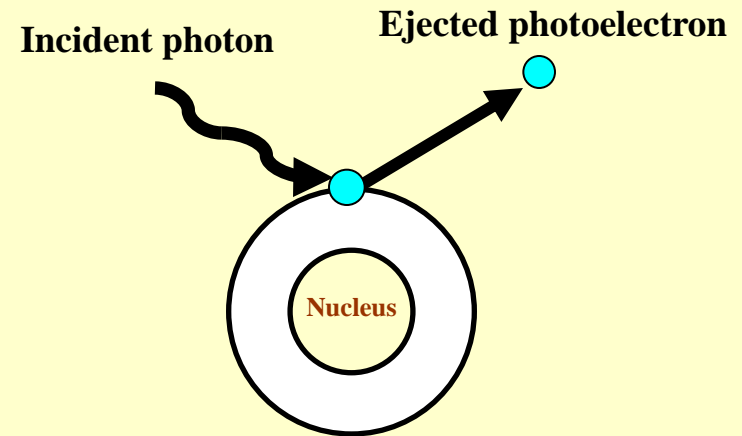
Interaction of photons with matter

■ The most important types of interaction ■

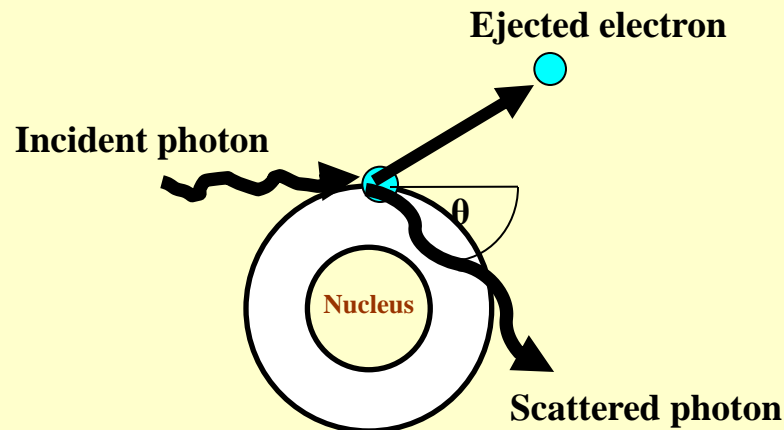
1) Rayleigh scattering (σ_{coh})



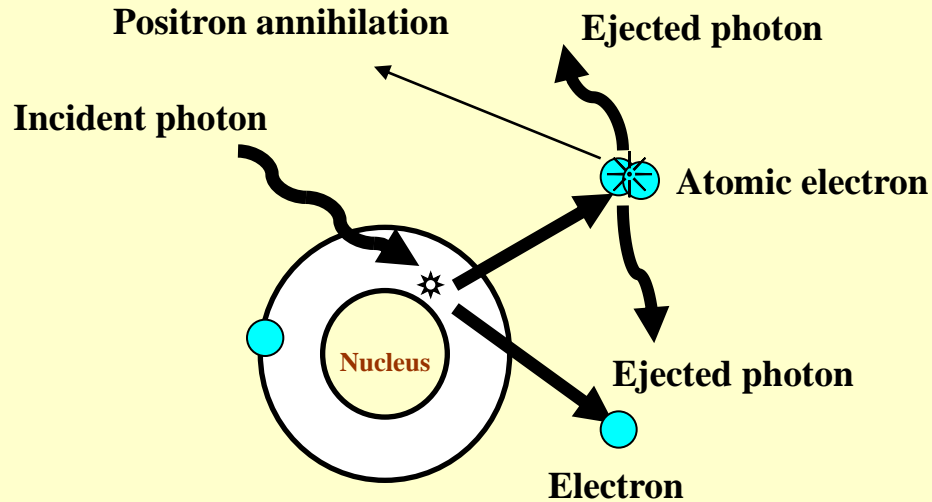
2) Photoelectric absorption (σ_{pe})



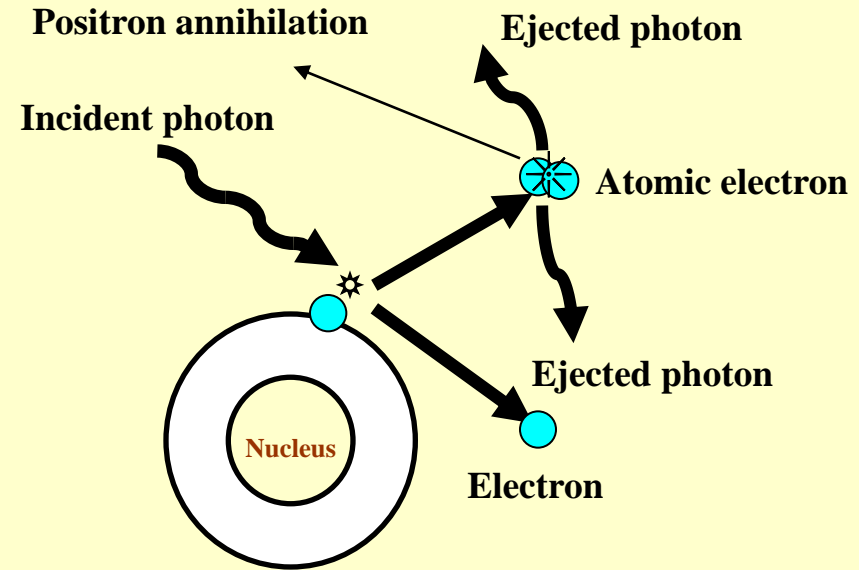
3) Compton scattering (σ_{incoh})



4) Pair production (σ_{pair})



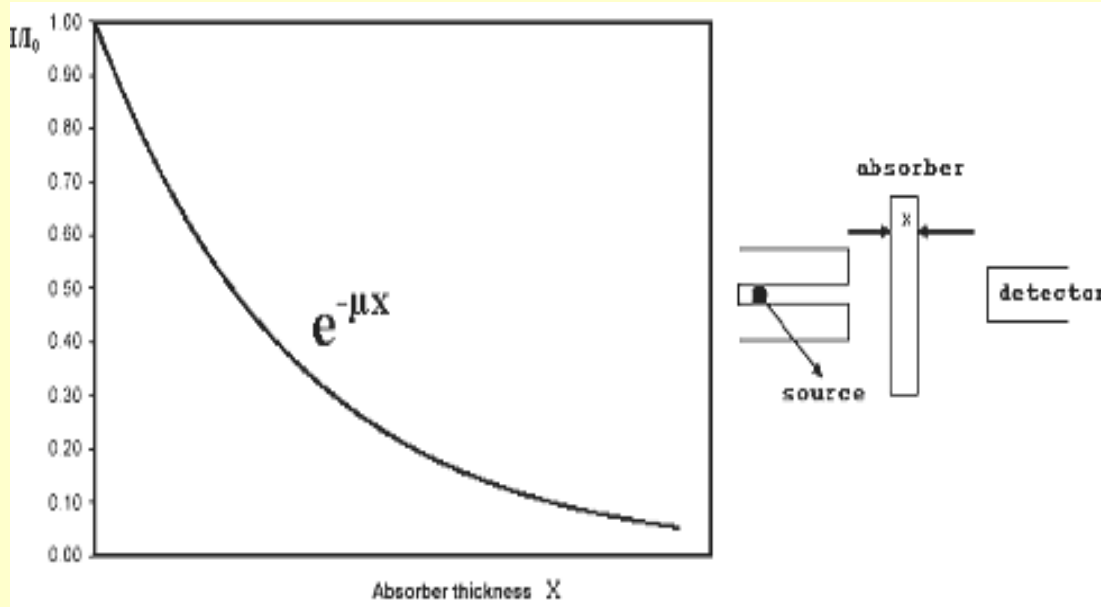
5) Triple production (σ_{triple})



The Total Attenuation Coefficients (μ)

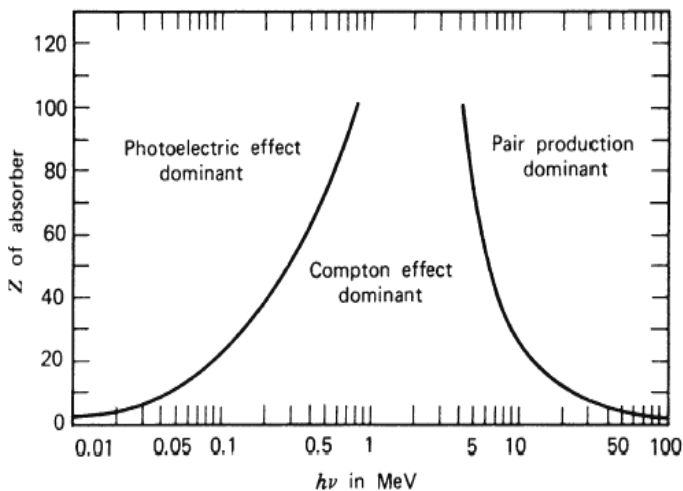
$$\mu = \sigma_{\text{coh}} + \sigma_{\text{pe}} + \sigma_{\text{incoh}} + \sigma_{\text{pair}} + \sigma_{\text{triple}}$$

Gamma-ray Attenuation

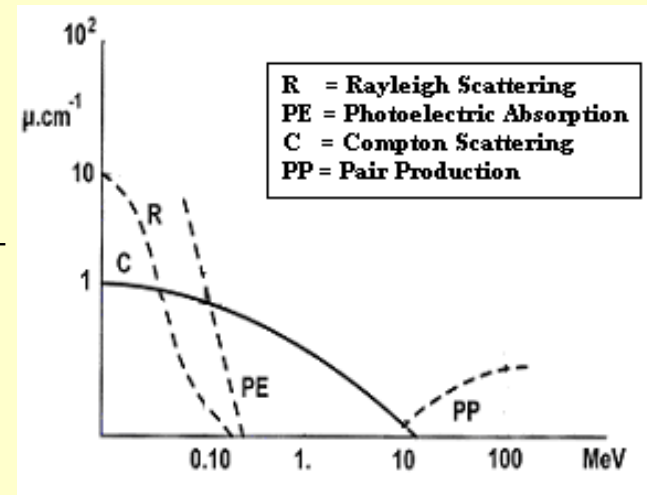


$$I = I_0 e^{-\mu X}$$

$$I = I_0 e^{-\left(\frac{\mu}{\rho}\right) \rho X}$$



$$\frac{1}{\mu} = \lambda = \frac{\int_0^{\infty} x e^{-\mu x} dx}{\int_0^{\infty} e^{-\mu x} dx}$$

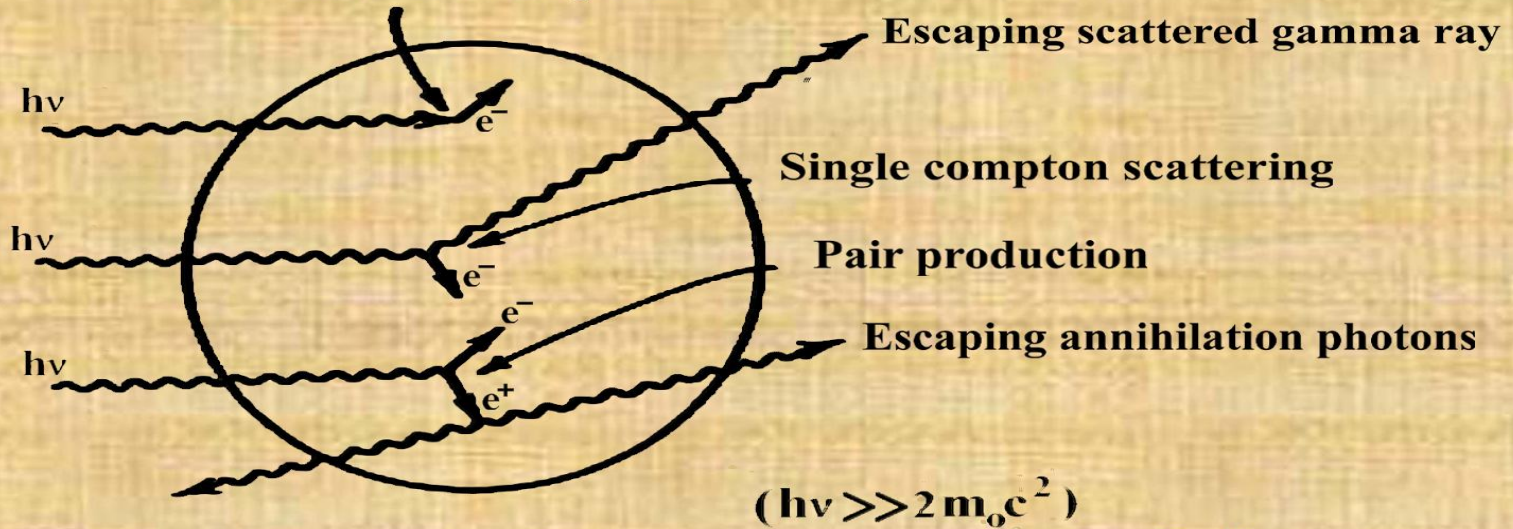


Energy deposition and spectrum for various detector sizes

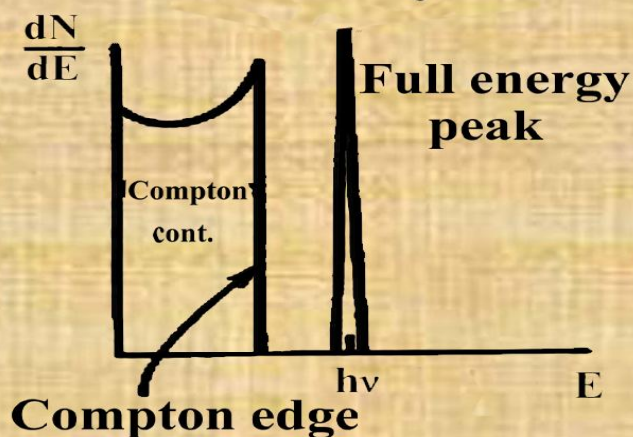


Small detectors

Photoelectric absorption

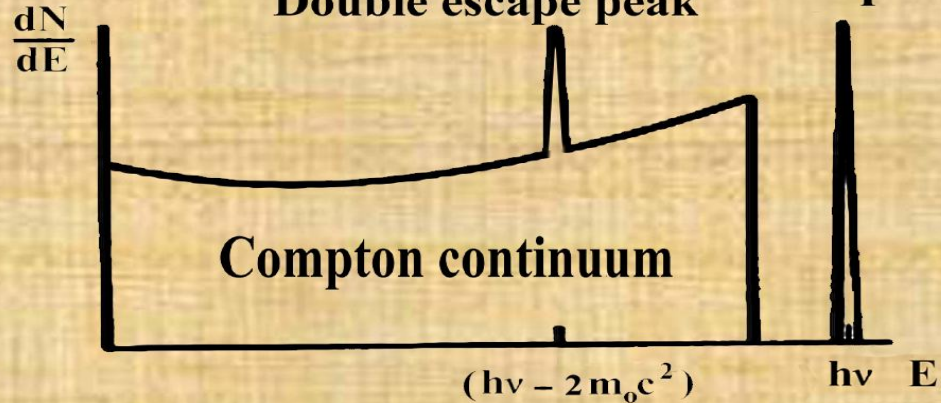


($h\nu < 2m_0c^2$)



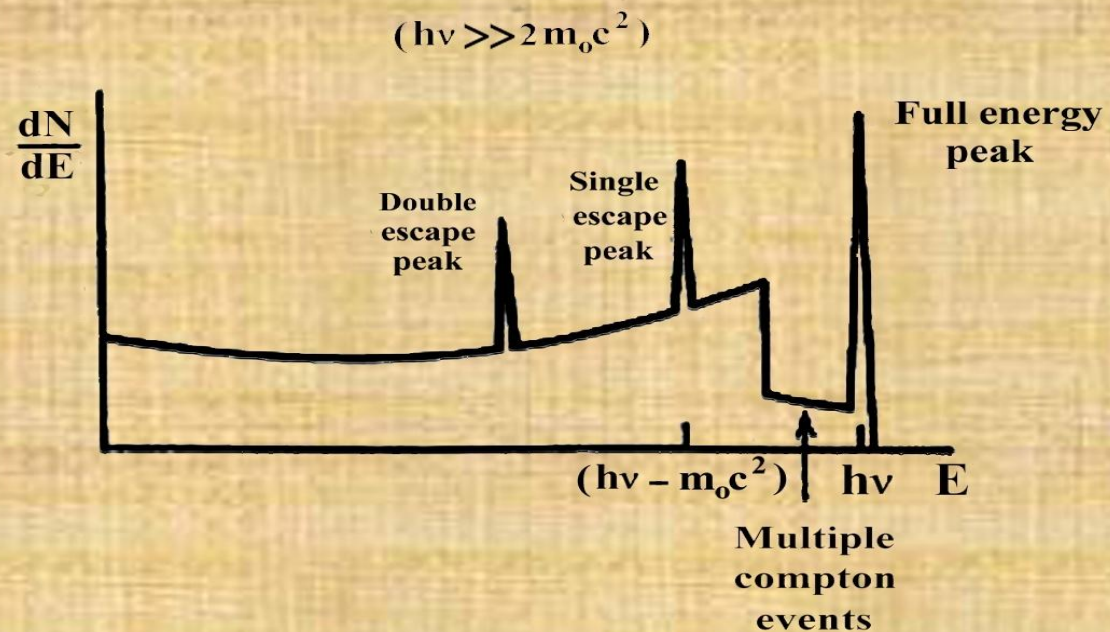
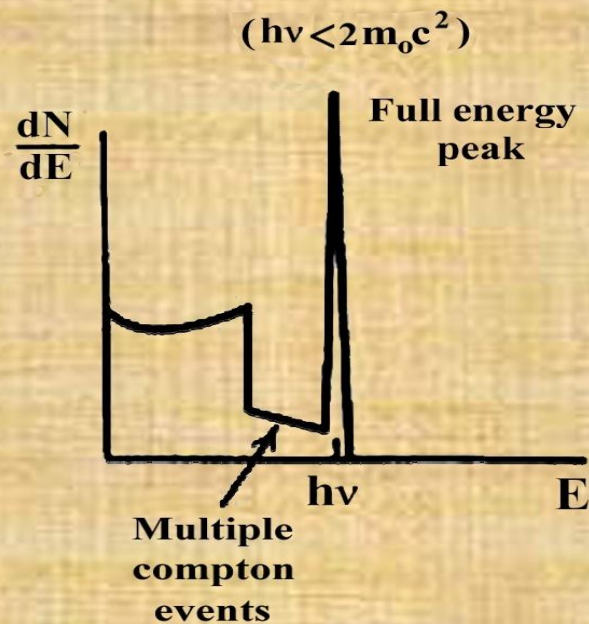
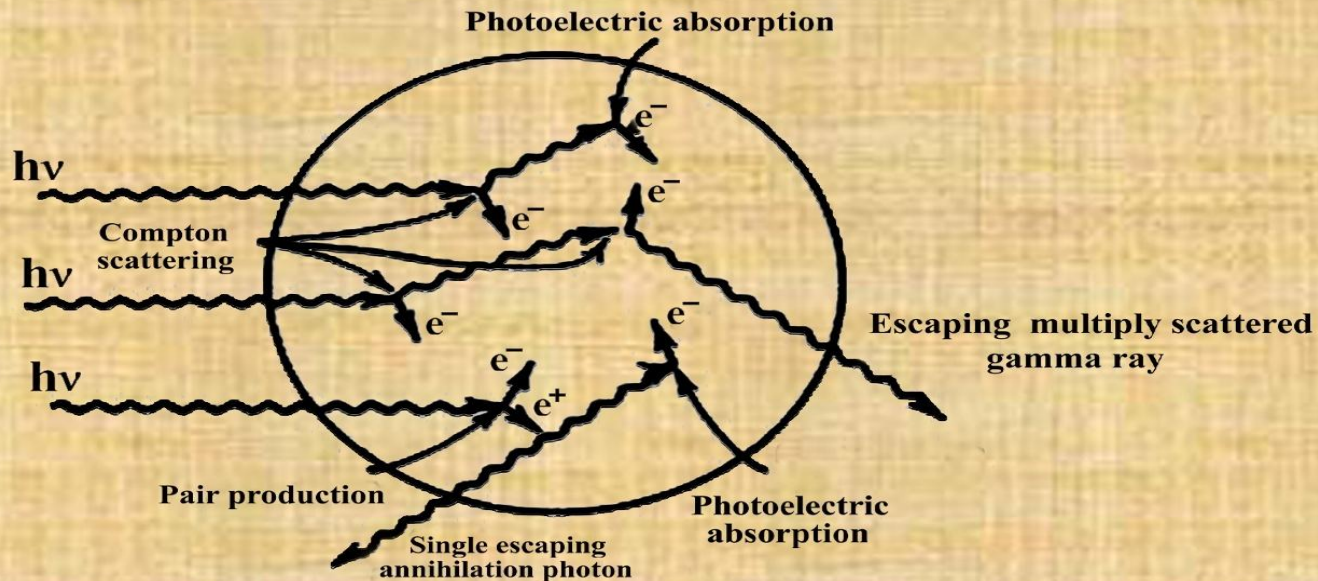
Full energy peak

Double escape peak



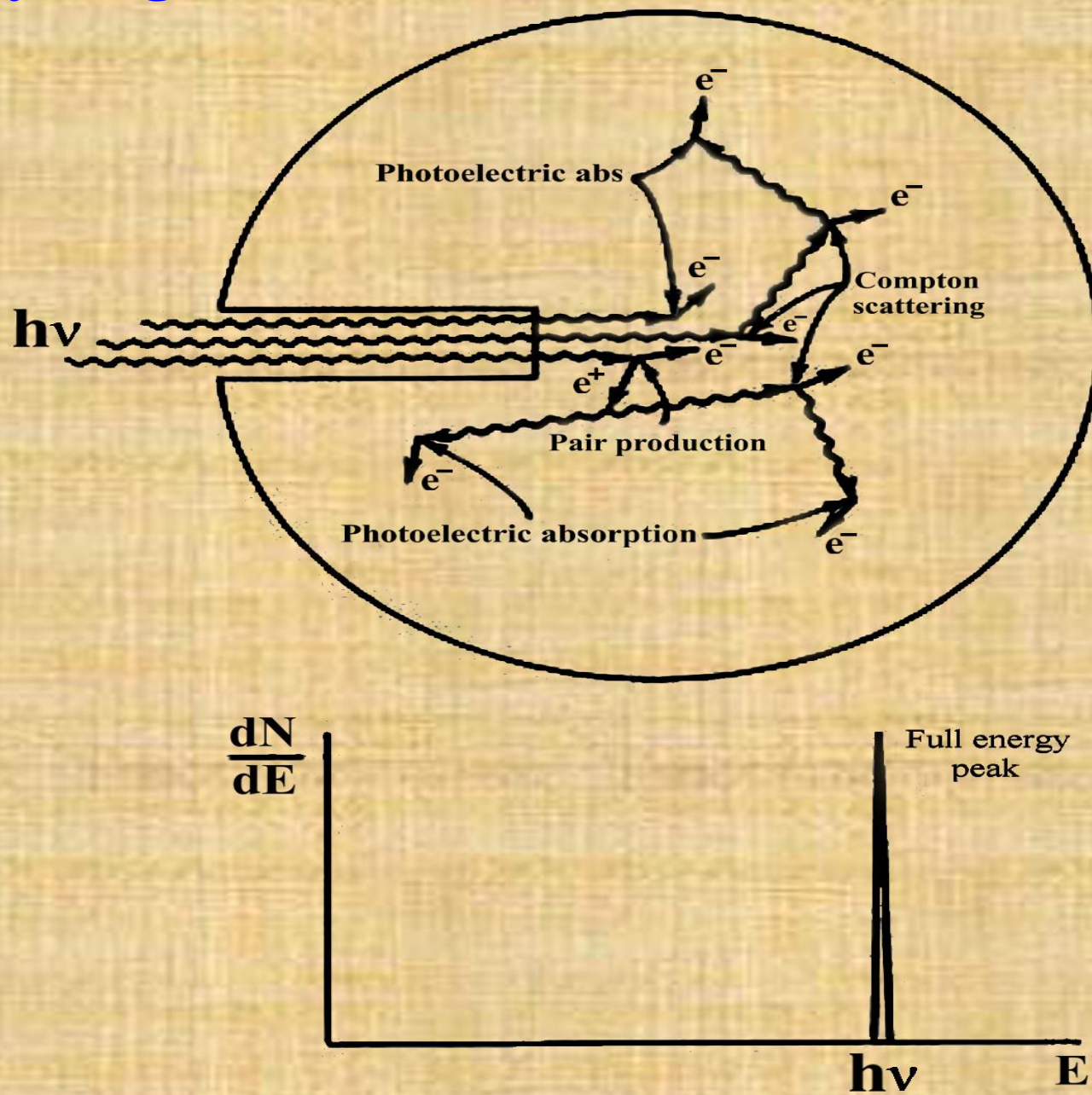


Intermediate size detectors



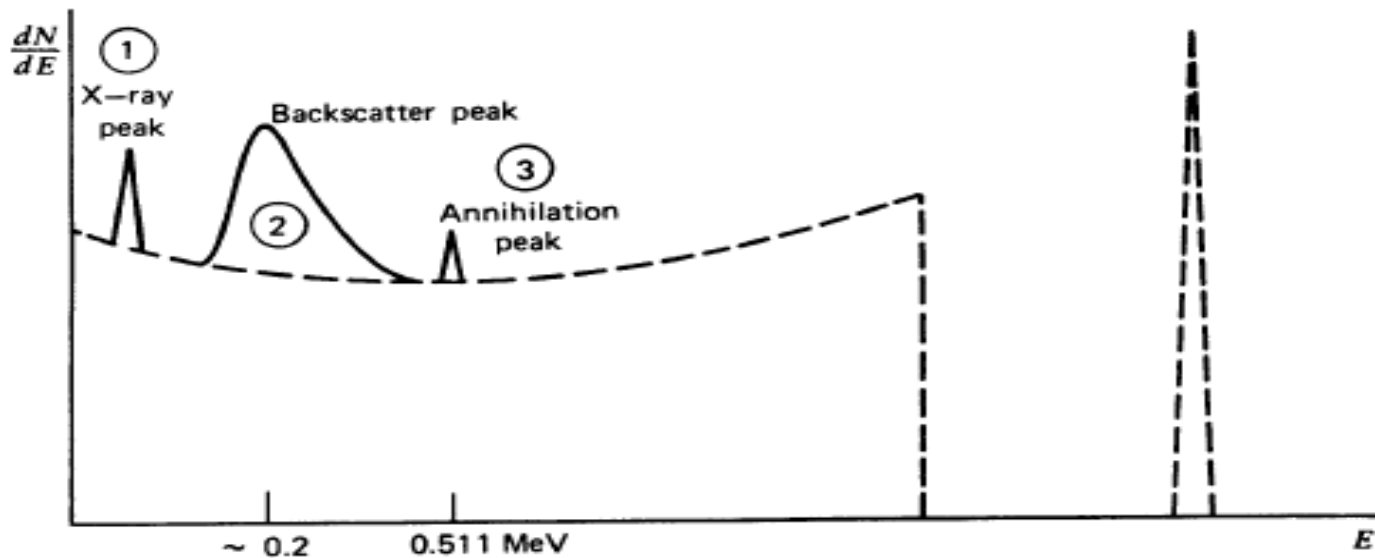
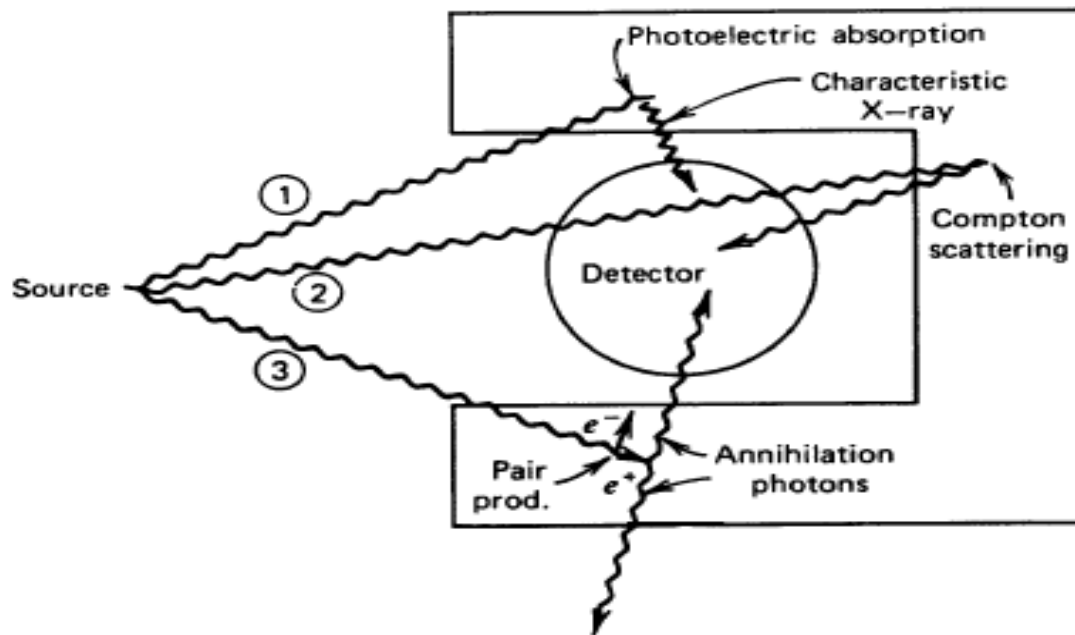


Very large detectors





Effects of surrounding materials



Modes of energy deposition in the detector.

Energy deposition by photons with $E_\gamma < 1.022 \text{ MeV}$ & $E_\gamma > 1.022 \text{ MeV}$

- The full energy peak.

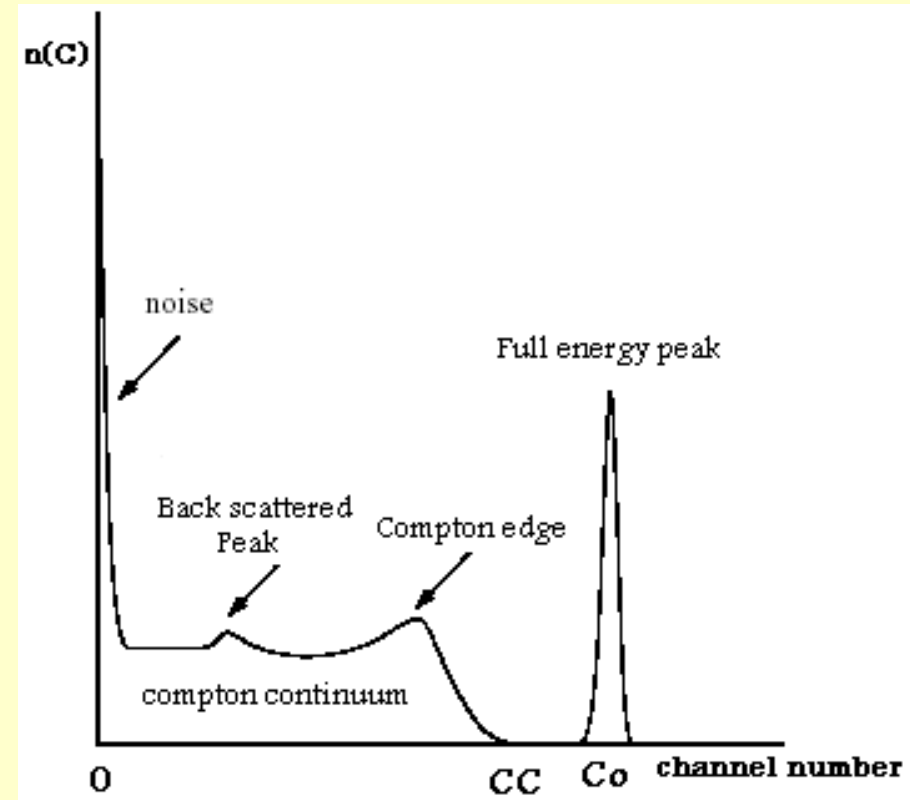
- The Compton edge.

Other peaks may be observed are:

- Backscatter peak.

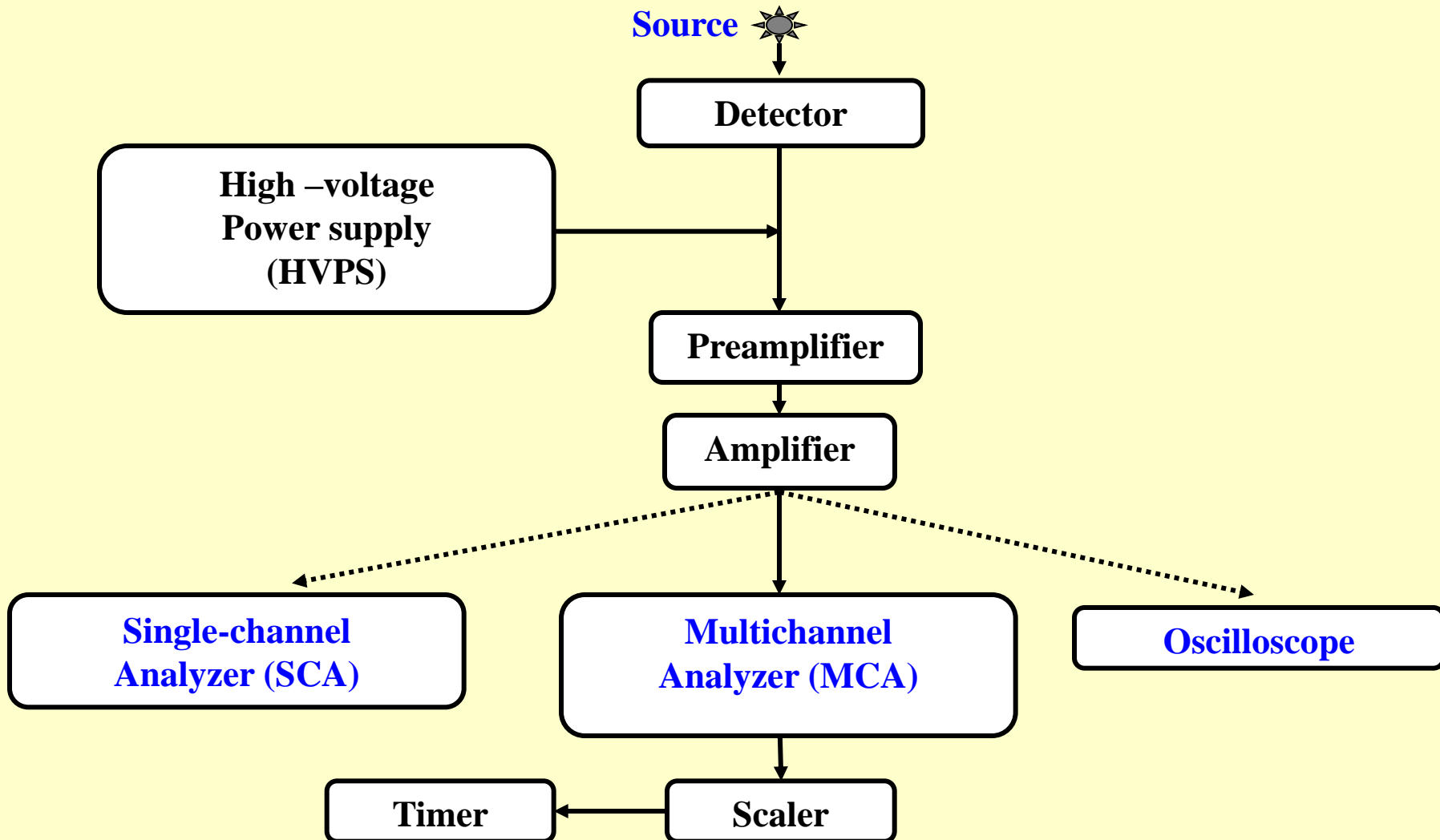
- The single-escape peak.

- The double-escape peak.



Detection and Recording System

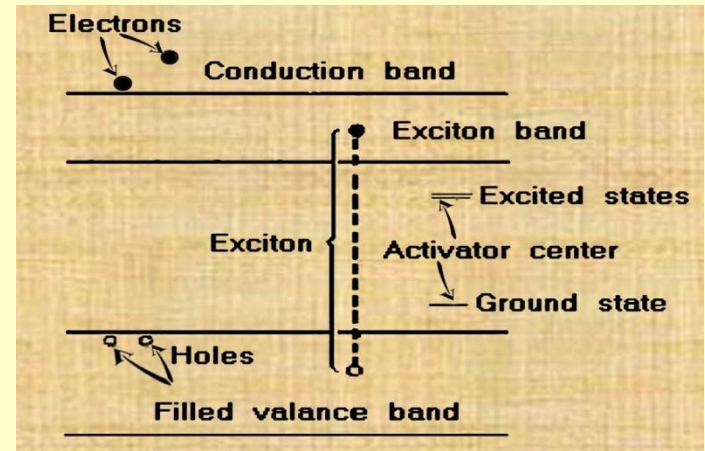
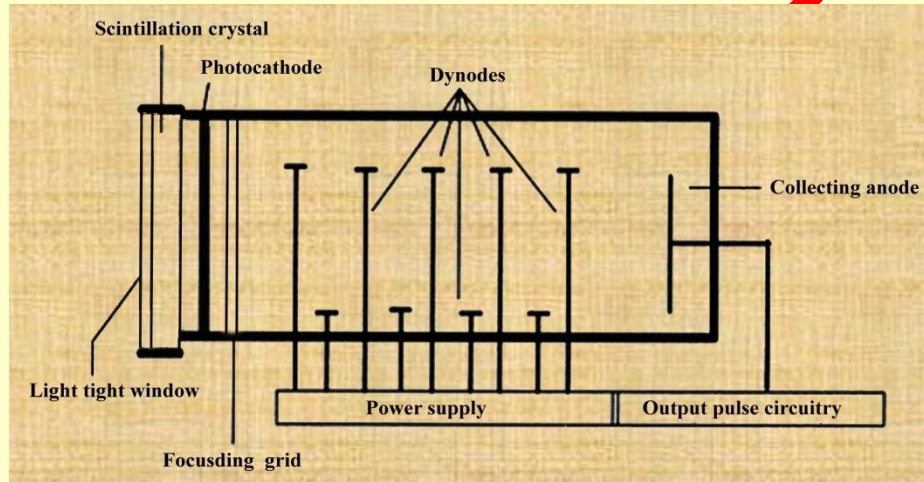
“ Method of Measurements”



Typical gamma detector.

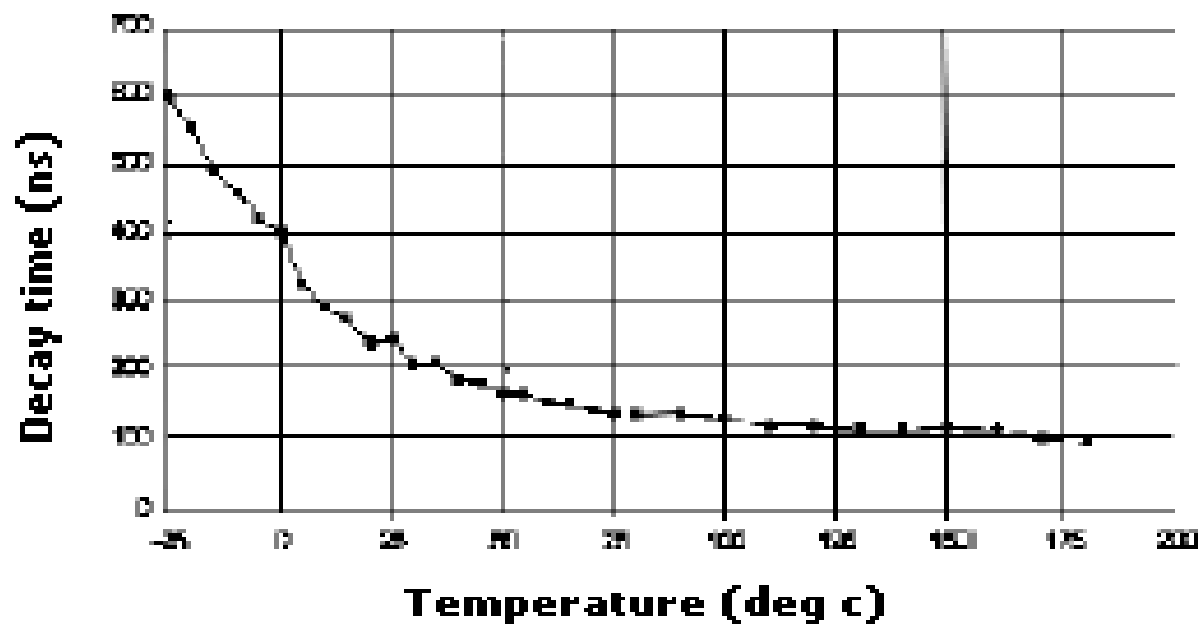
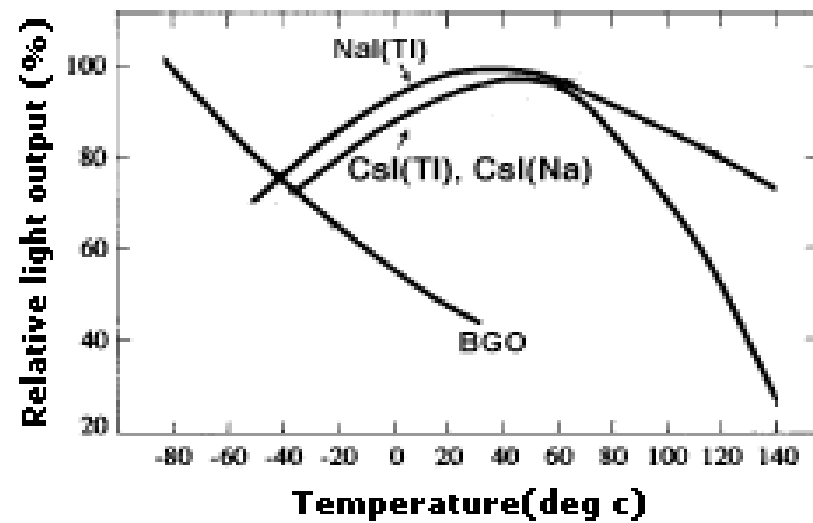
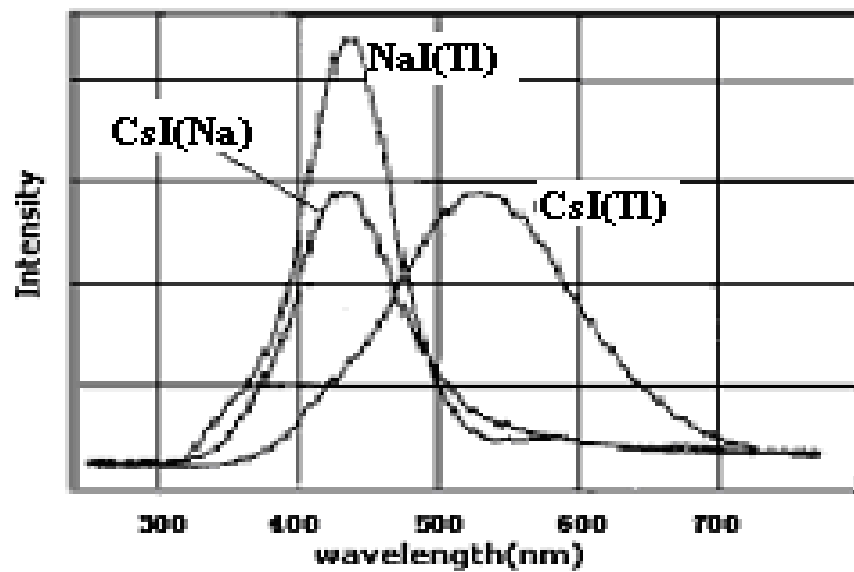
Types of detectors

1) Scintillation detectors → The NaI(Tl) scintillation detector.



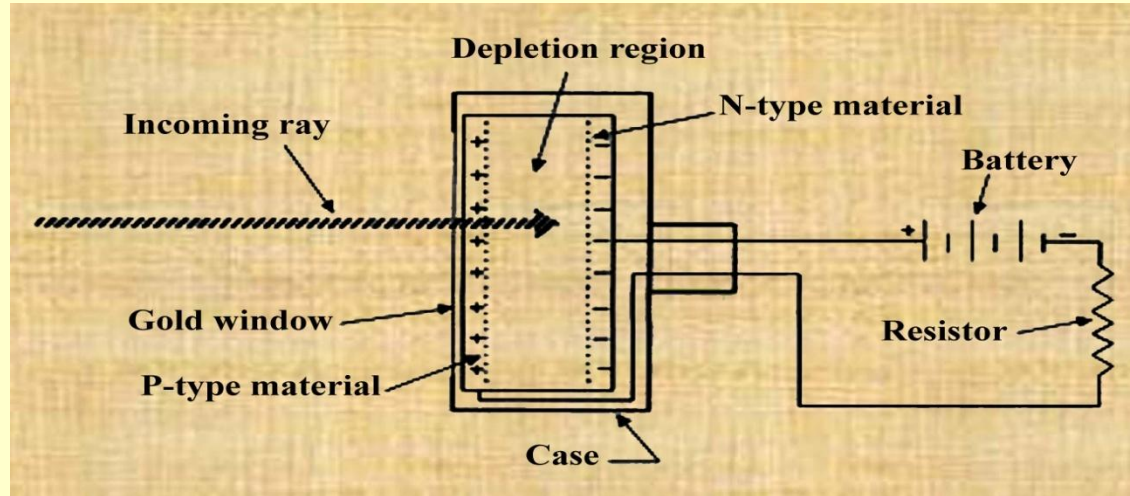
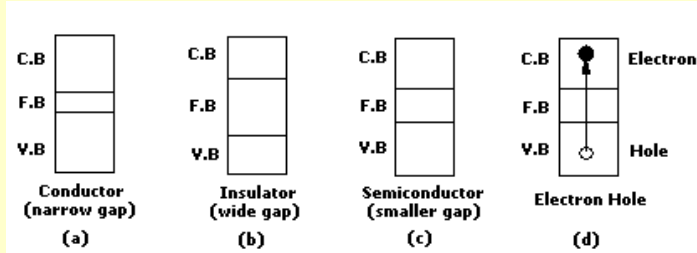
↙ The mechanism of the scintillation process ↘

- Ionizing radiation passes through the crystal.
- Electrons are raised to the conduction band.
- Holes are created in the valance band.
- Excitations are formed.
- Activation centers are raised to the excited states by absorbing electrons, holes and excitons.
- De-excitation followed by the emission of photons.
- The emission photons interact with the photocathode and electrons are emitted.
- The electrons are multiplied by dynodes in photomultiplier tube and finally produced the electric pulse.



2) Semiconductor detectors

- ➔ Germanium lithium detector (GeLi).
- ➔ Hyper pure germanium detector (HPGe).



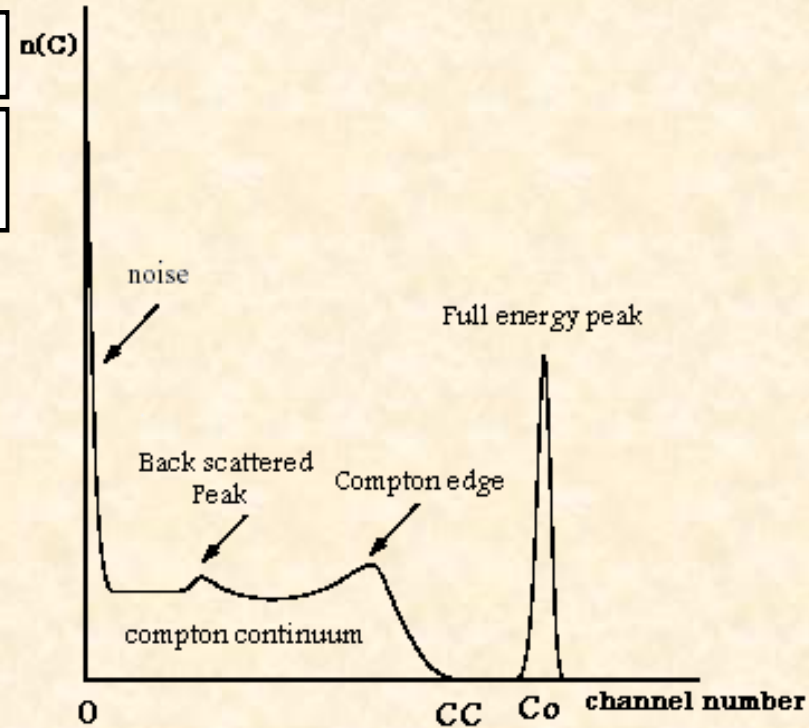
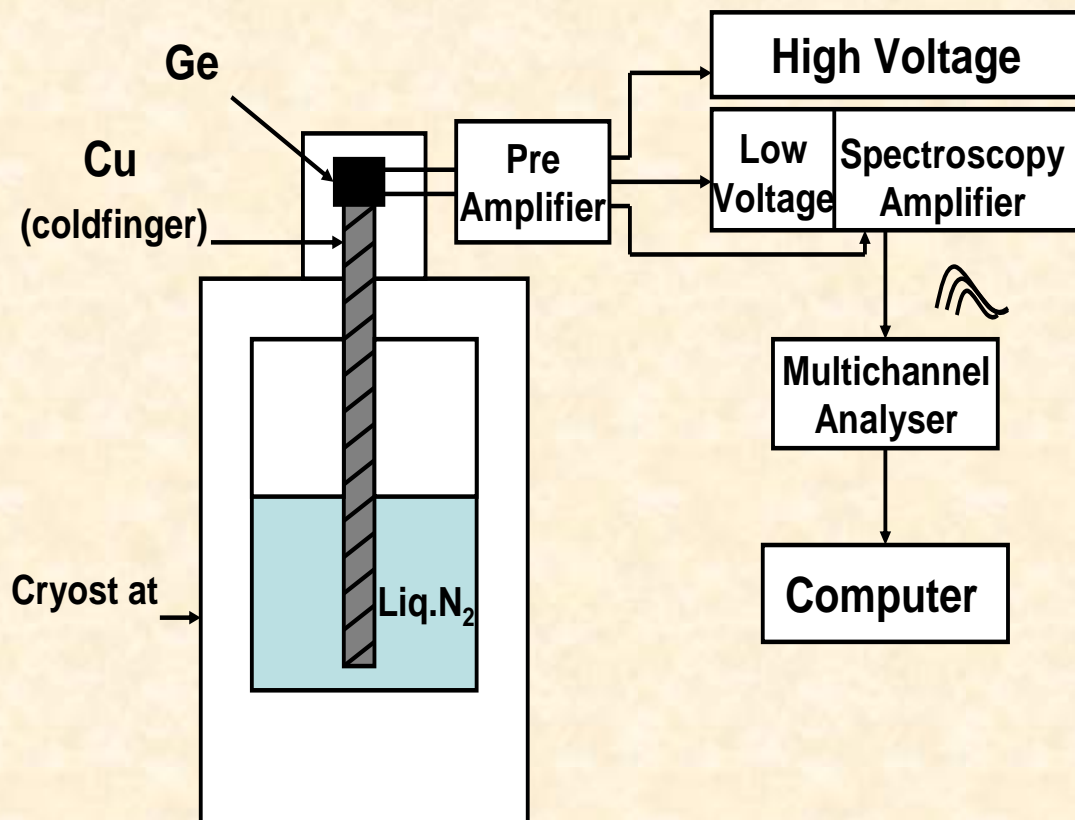
↪ The mechanism of the junction counter ↪

- Put a suitable external electric field (reverse bias).
- Formed the depletion region.
- Incoming ray interaction.
- Creating electron-hole pairs.
- Collection of the electron-hole pairs to produce the electric pulse.

Detection and recording system

Semiconductor Detectors

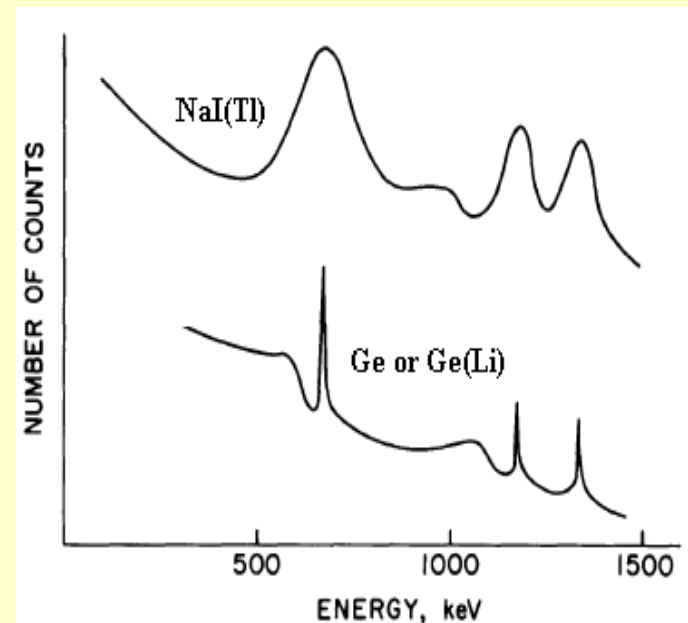
Complete HPGe spectrometer



Spectrum Energy Distribution For Photons

➤ Advantages of Ge detectors.

- High resolution & low efficiency.
- Short response time.
- Linear energy response.
- Small crystals size.
- Rise to room temperature.
- don't need liquid nitrogen.
- Portable detector.
- Very low resistance.

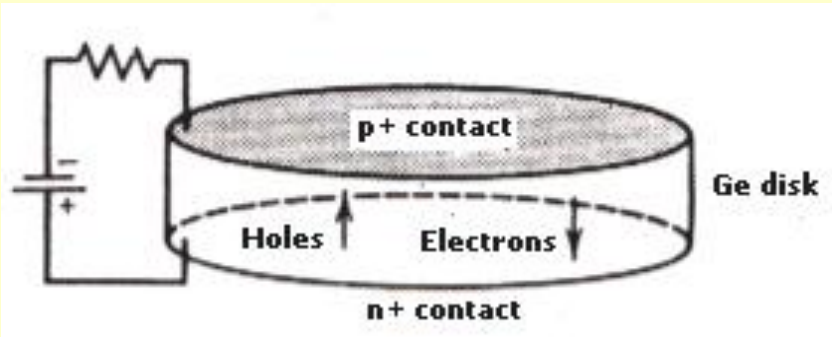


➤ Disadvantages of Ge detectors.

- Used for gamma photon only.
- Need liquid nitrogen.
- Very expensive.
- Small crystals size.
- Good for environmental samples.
- don't need liquid nitrogen.
- Portable detector.



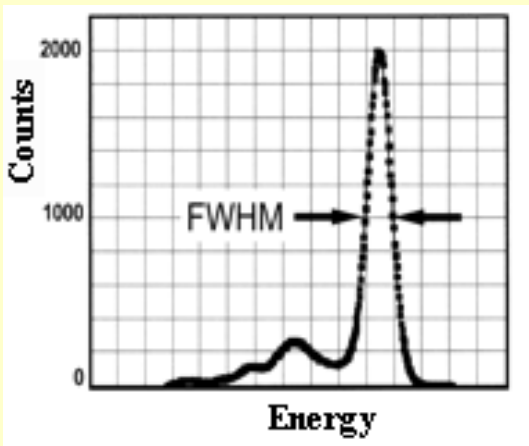
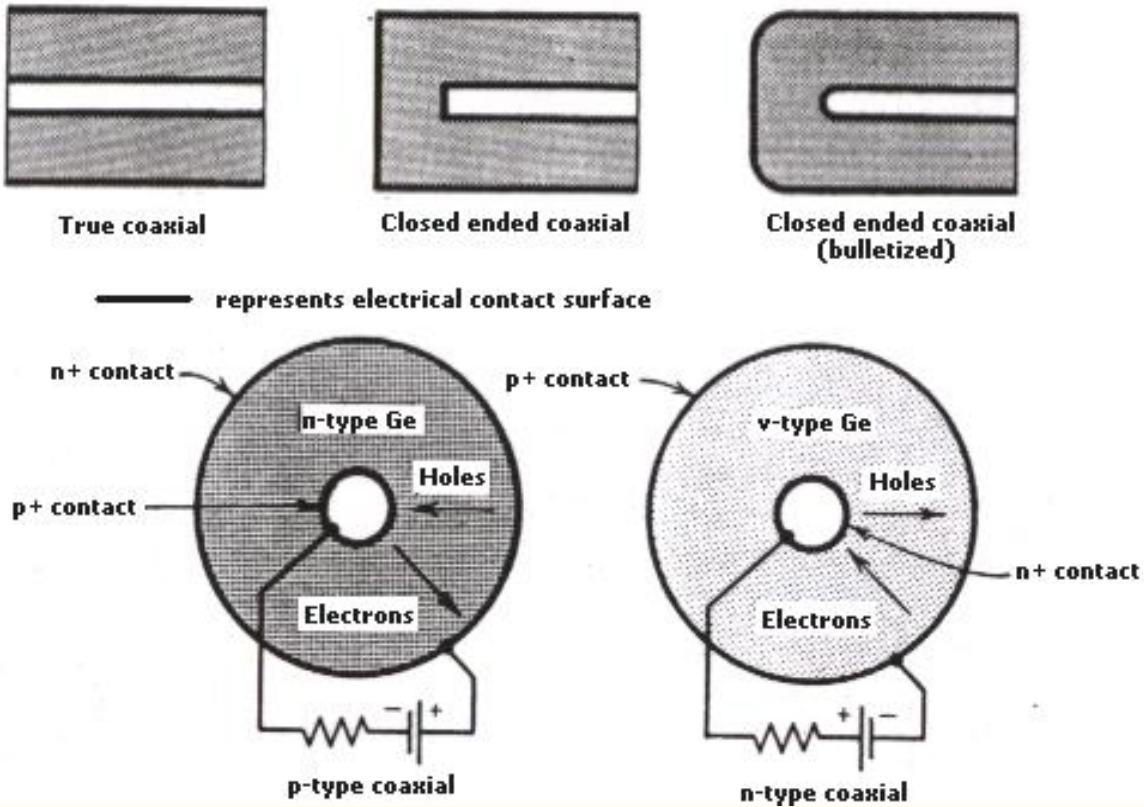
Planar detector



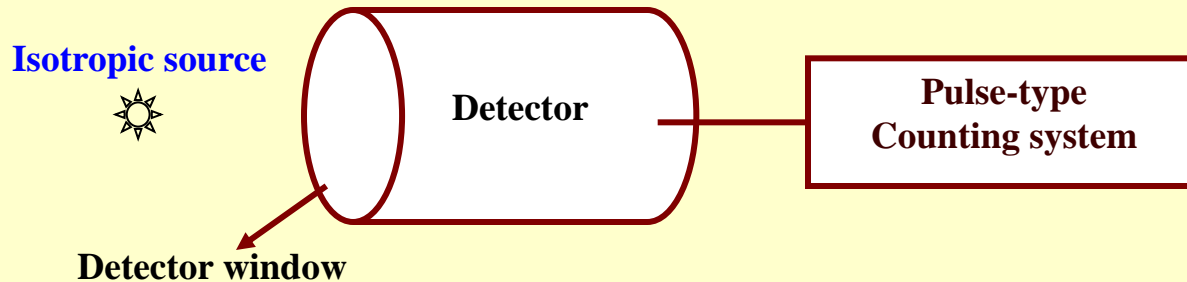
Energy resolution

$$\text{Energy resolution} = \frac{\text{FWHM}}{\text{PE}} \times 100\%$$

Coaxial detector



Detectors factors



Let

N = number of the particles per second **emitted** by the source.

r = number of particles per second **recorded** by the scaler.

The **counting rate r** is related to **N** by :

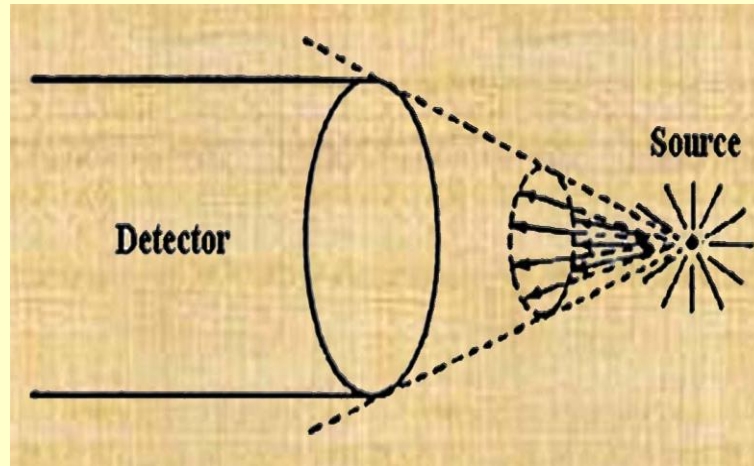
$$r = N \cdot f_1 f_2 f_3 \dots f_n$$

Where the *f* factors represent the effects of the experimental setup on the measurements. These factors may be grouped into four categories.

① Geometry effects.

- ➡ Size and shapes of source (point,disk,rectangular,etc).
- ➡ Size and shapes of detector (cylindrical,rectangular,etc).
- ➡ Source-to-detector distance.

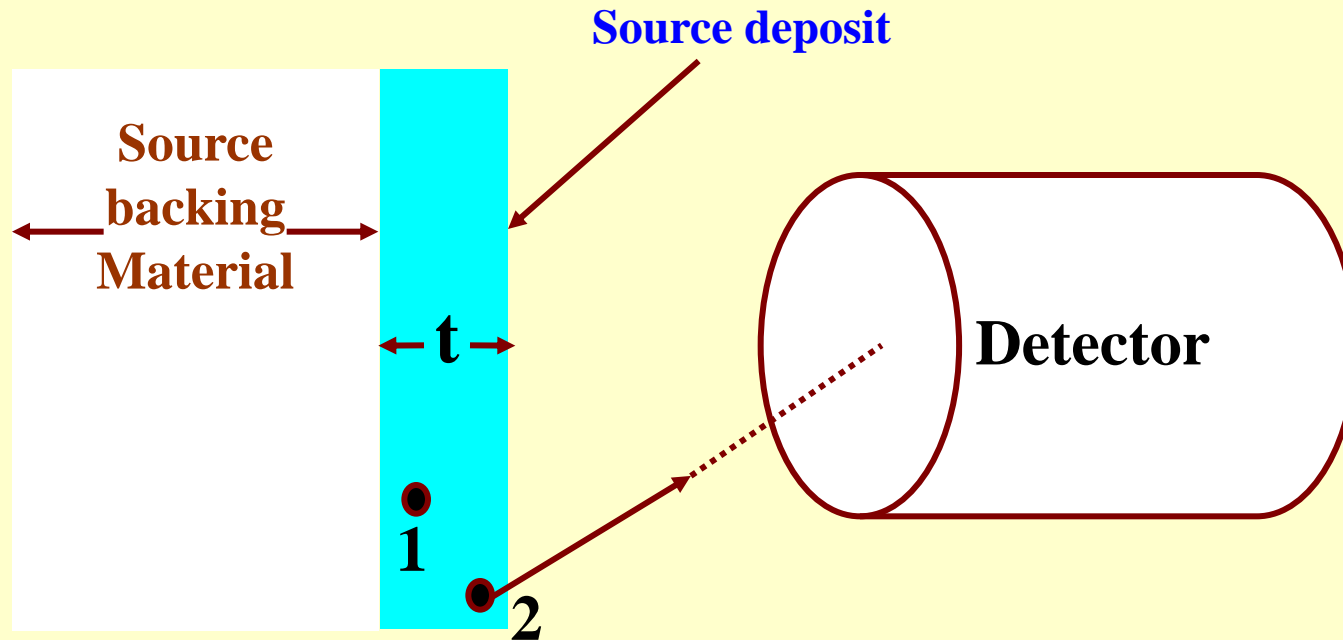
All these factors are included in the **solid angle (Ω)**



$$\Omega = \frac{\text{number of photons that enter the detector}}{\text{number of photons that are emitted from the source}}$$

② source effects.

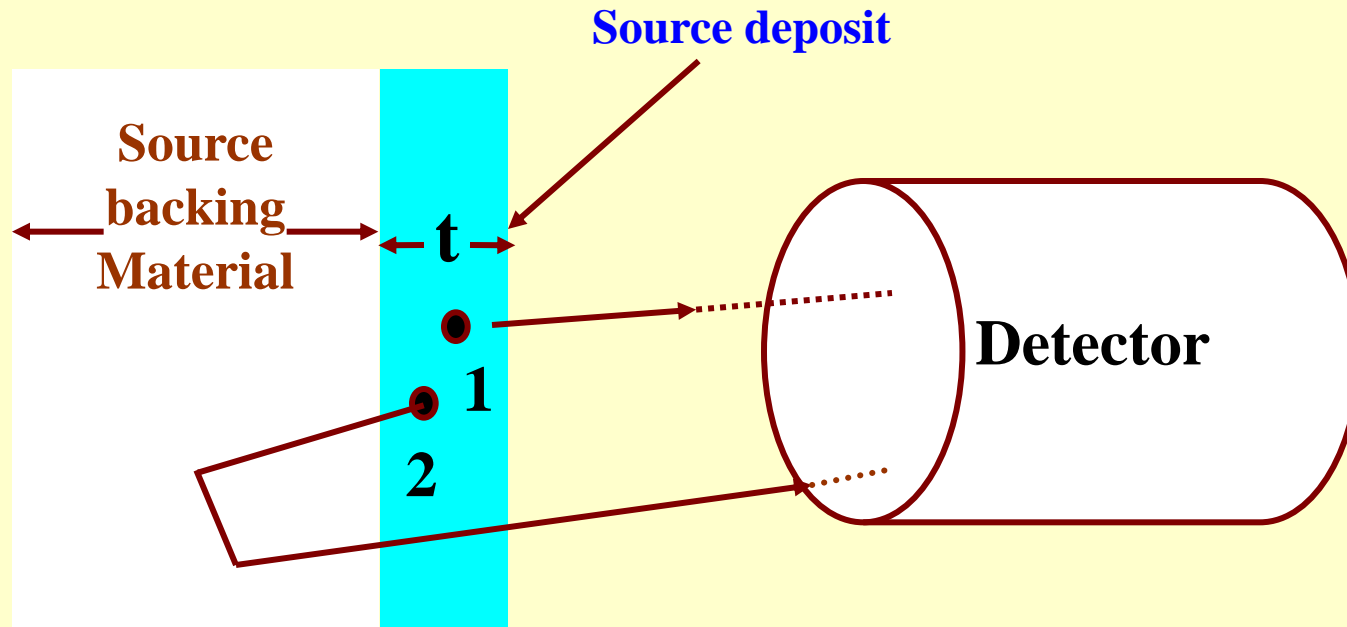
➡ Source self-absorption factor (f_{self})



A **Source self-absorption factor** (f_{self}) is defined by:

$$f_{\text{self}} = \frac{\text{number of particles leaving source with self - absorption}}{\text{number of particles leaving source without self - absorption}}$$

► Source backscattering factor (f_{back})

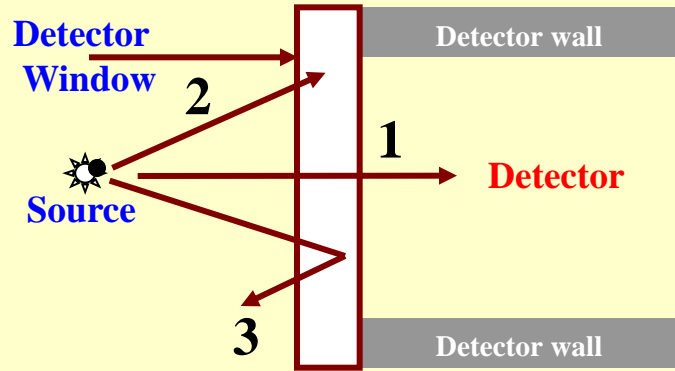


A **Source backscattering factor** (f_{back}) is defined by:

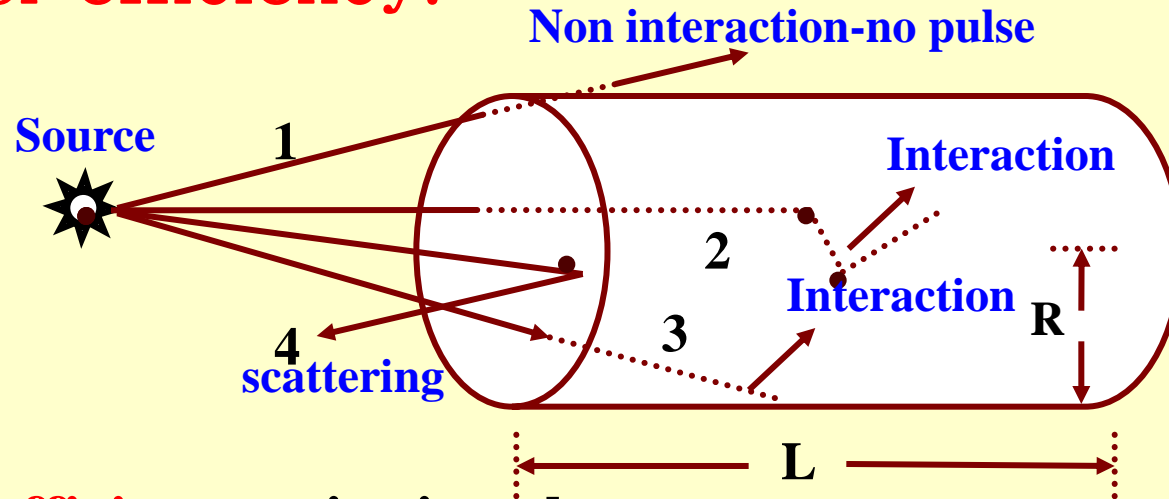
$$f_{back} = \frac{\text{number of particles counted with source backing}}{\text{number of particles counted without source backing}}$$

③ The medium between source and detector.

➡ Thickness of the detector **window** or **any absorber** which cause **attenuation**.



④ Detector efficiency.



The **detector efficiency** ϵ is given by:

$$\epsilon = \frac{\text{number of particles recorded per unit time}}{\text{number of particles impinging upon the detector per unit time}}$$

Types of the detector efficiency

1) The geometrical efficiency (ϵ_g).

“The ratio between the number of photons that are **enter** the detector and the number of photons that are **emitted** from the source”.

$$\epsilon_g = \frac{\Omega}{4\pi}$$

2) The intrinsic efficiency (ϵ_{iT}).

“The ratio between the number of photons that are **recorded** in the detector and the number of photons that are **enter** the detector”.

3) The total efficiency (ϵ_T).

“The ratio between the number of photons that are **recorded** in the detector with any possible energy during a certain time interval and the number of photons that are **emitted** by the source during the same time interval”.

$$\epsilon_T = \epsilon_g \cdot \epsilon_{iT}$$

4) The intrinsic photopeak efficiency (ϵ_{ip}).

“The ratio between the number of photons that are **recorded** under a certain energy peak and the number of photons that are **enter** the detector with the energy relates to this peak”.

5) The full energy peak efficiency (FEPE)(ϵ_p).

“The ratio between the number of photons that are **recorded** under a certain energy peak and the number of photons that are **emitted** from the source with the energy relates to this peak”.

$$\epsilon_p = \epsilon_g \cdot \epsilon_{ip}$$

6) The single escape peak efficiency (ϵ_{se}).

“The ratio between the number of **single escape photons** and the number of photons that are **emitted** from the source”.

7) The double escape peak efficiency (ϵ_{de}).

“The ratio between the number of **couple photons escape** and the number of photons that are **emitted** from the source”.

8) The photo-fraction (**peak-to-total ratio**)(**p**).

“The ratio between the number of photons that are **recorded** under a certain energy peak and the number of photons that are **recorded** in all the spectrum at the same energy”.

$$P = \frac{\varepsilon_{ip}}{\varepsilon_{iT}} = \frac{\varepsilon_p}{\varepsilon_T}$$

9) The relative efficiency.

$$(\text{Relative efficiency})_i = \frac{(\text{The efficiency of the detector})_i}{\text{Efficiency of the standard detector}}$$

Where the **subscript i** refers to any one of the efficiencies defined before. (The standard detector is a **3“x 3 “ NaI(Tl)** detector. The efficiency is measured for photon with energy **1.33 MeV** and **source-to-detector distance 25 cm**)



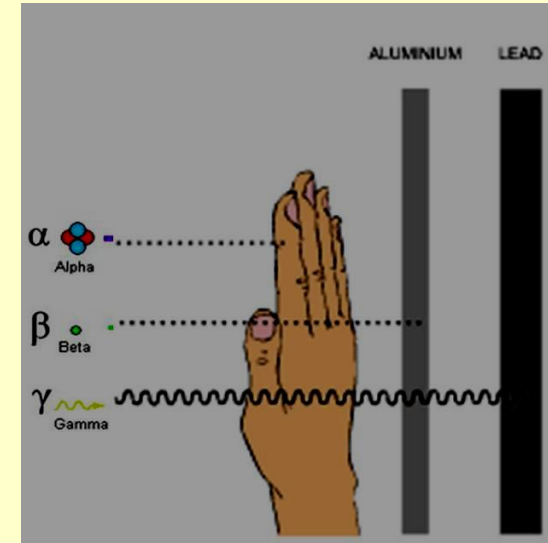
Alpha Particles

➡ The alpha particle is a **helium nucleus (2 protons, 2 neutrons)** produced from the radioactive decay of heavy metals and some nuclear reactions.

➡ The **high positive charge (2+)** of an alpha particle causes **electrical excitation and ionization of surrounding atoms**.

➡ Alpha particles are the **least penetrating** radiation featuring a relatively straight path over a **short distance (several cm in air)**.

➡ The **specific ionization** of alpha particles is very high.



Beta Particles

- There are two types of beta particles: electron (β^-) and positron (β^+). These are ejected from the nucleus of a beta-unstable radioactive atom .
- The beta has a single negative or positive electrical charge and **very small mass**.
- The interaction of a beta particle and an orbital electron leads to electrical **excitation and ionization of the orbital electron**.
- Beta particles follow a tortuous path (**zig-zag**).
- The **specific ionization** of a beta particle is **low** due to its **small mass, small charge, and relatively high speed of travel**.
- The interaction of β^+ (positron) with an electron leads to their **annihilation** and occurrence of two annihilation photons with the energy of 0.511 MeV.

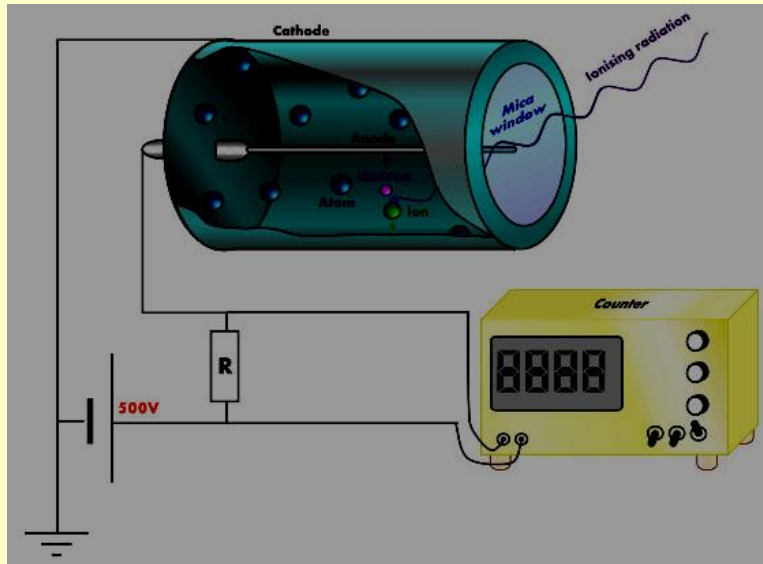
⇒ Neutrons

⇒ Neutrons have **no electrical charge** and have nearly the same **mass as a proton**.



⇒ Neutrons are fairly **difficult to stop**, and have a relatively **high penetrating power**.

Gas-Filled Detector

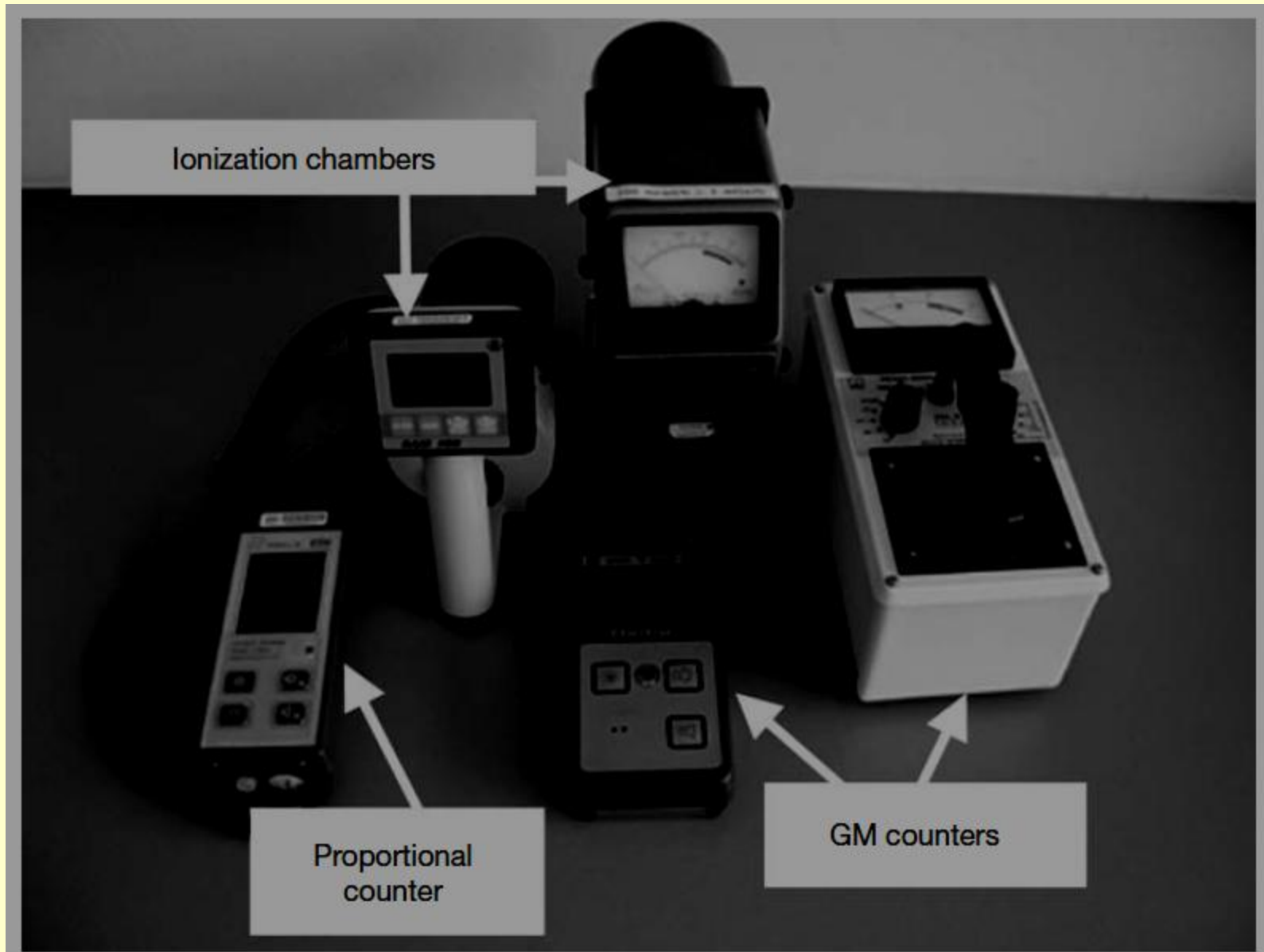


$$\Delta V = \Delta Q/C$$



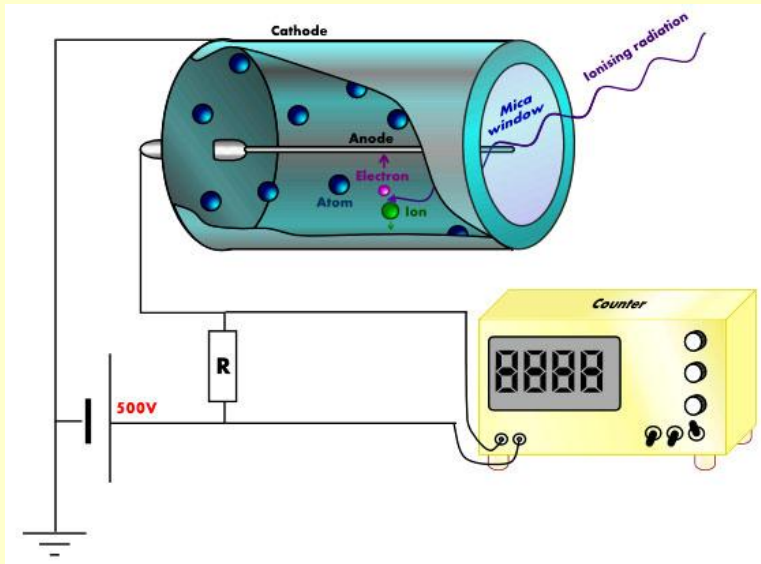
- The central electrode, or **anode**, attracts and collects the electron of the ion-pair.
- The chamber **walls** attract and collect the positive ion.
- When the applied voltage is high enough, the ion pairs initially formed accelerate to a high enough velocity to cause secondary ionizations. The resultant ions cause further ionizations. This multiplication of electrons is called **gas amplification**.

Gas Counters





Properties of Gas-Filled Detector



$$\Delta V = \Delta Q / C$$



- The **pulse height** can be computed if the **capacitance, detector characteristics, and radiation** are known.
- The **capacitance** is normally about 10^{-4} farads.
- The **number of ionizing events** may be calculated if the **detector size and specific ionization, or range of the charged particle, are known.**
- The **only variable** is the **gas amplification factor** that is **dependent on applied voltage.**



Detector Voltage

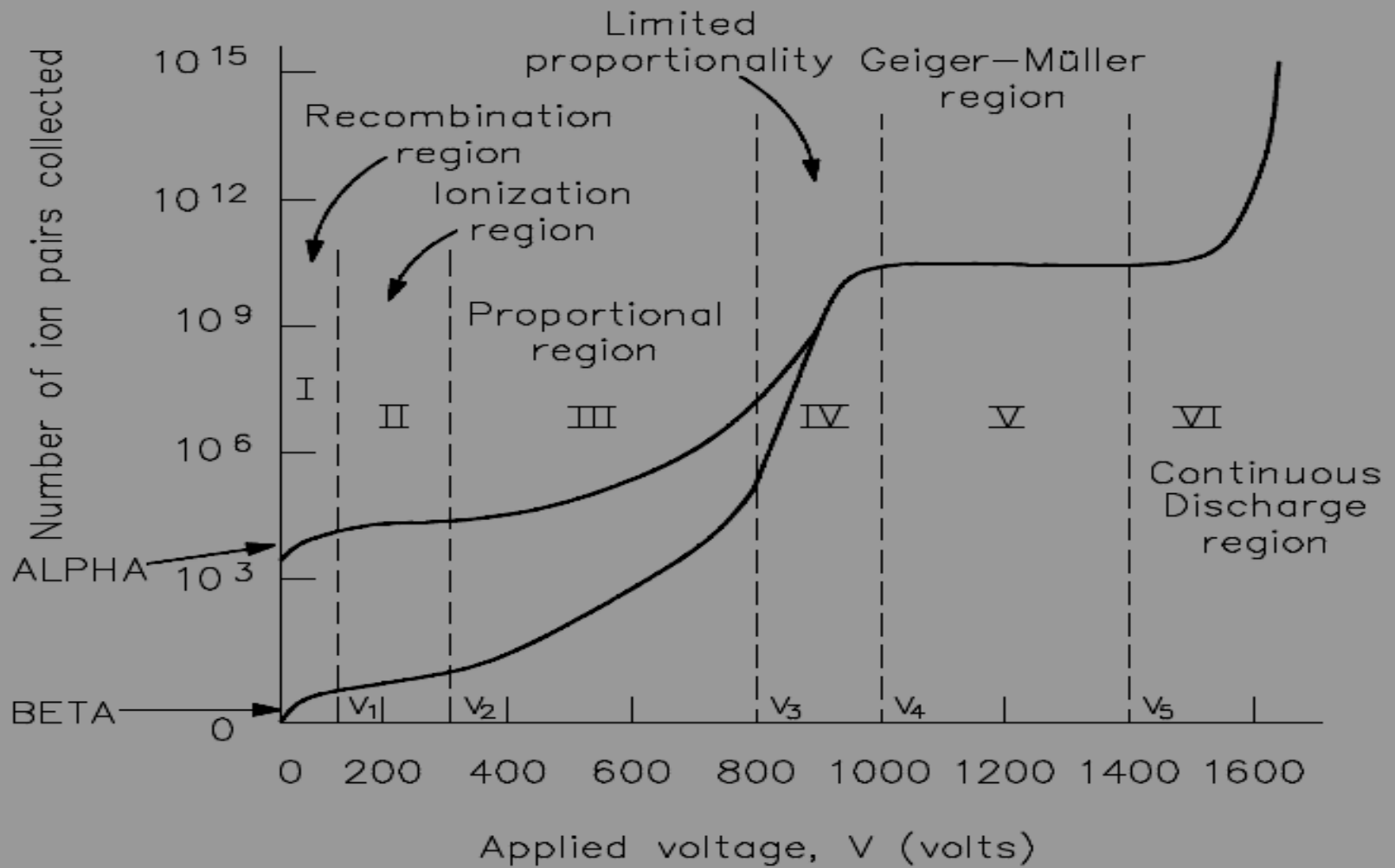
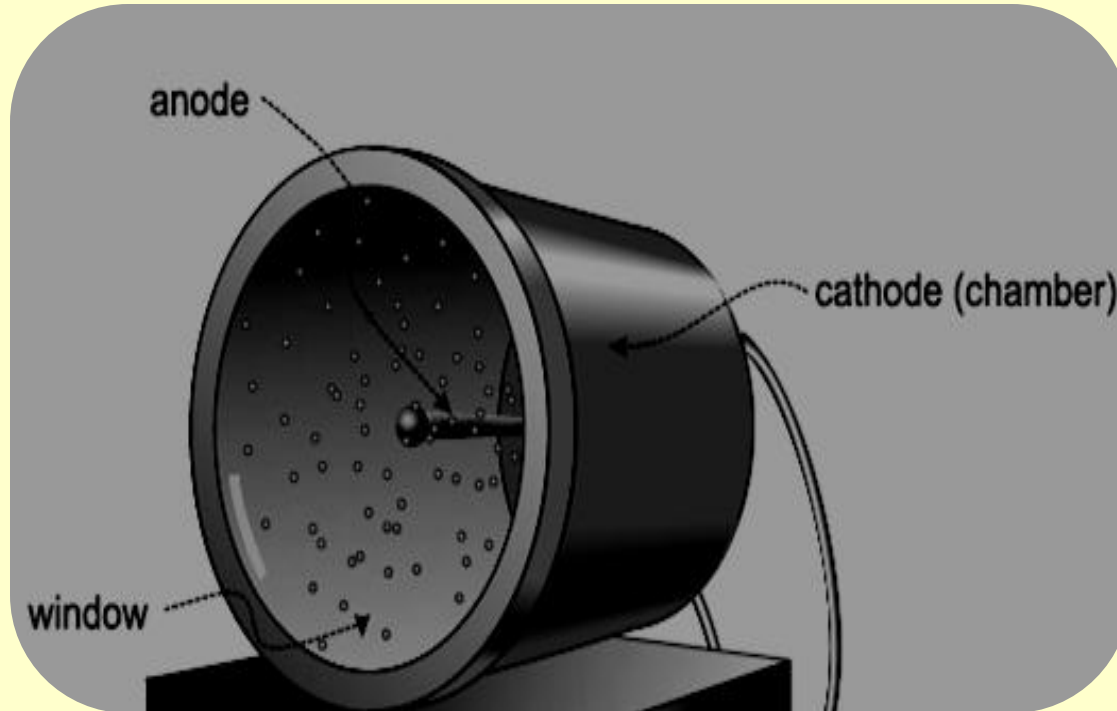


Figure 6 Ion Pairs Collected -vs- Applied Voltage



Detector Time

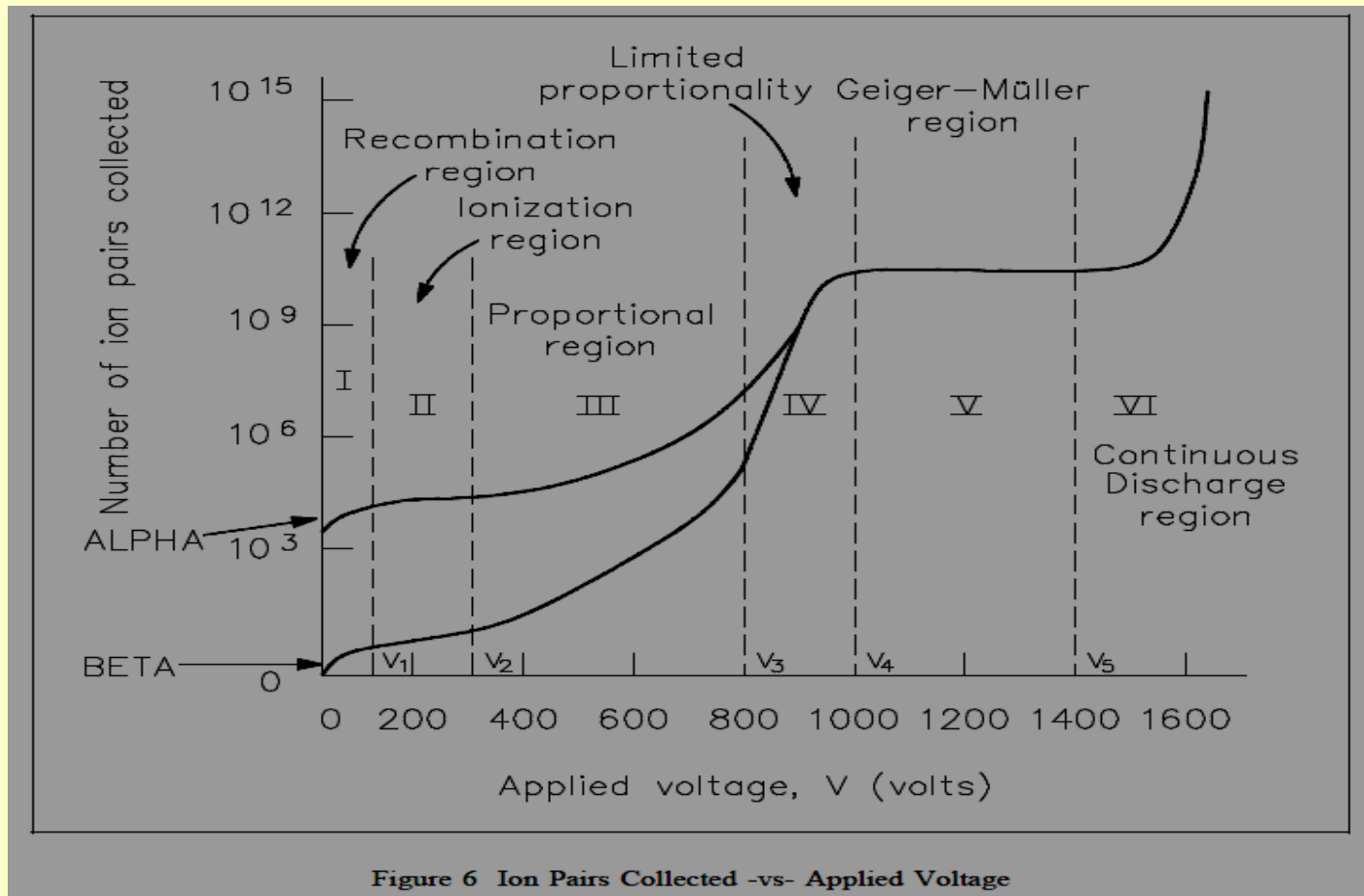


➡ **Resolving time** is the time required by the detector to regain its normal state after registering a pulse

➡ **Dead time** is the time during which the second pulse could not be detected, It is caused by the slow movement of the positive ions away from the anode



Recombination Region

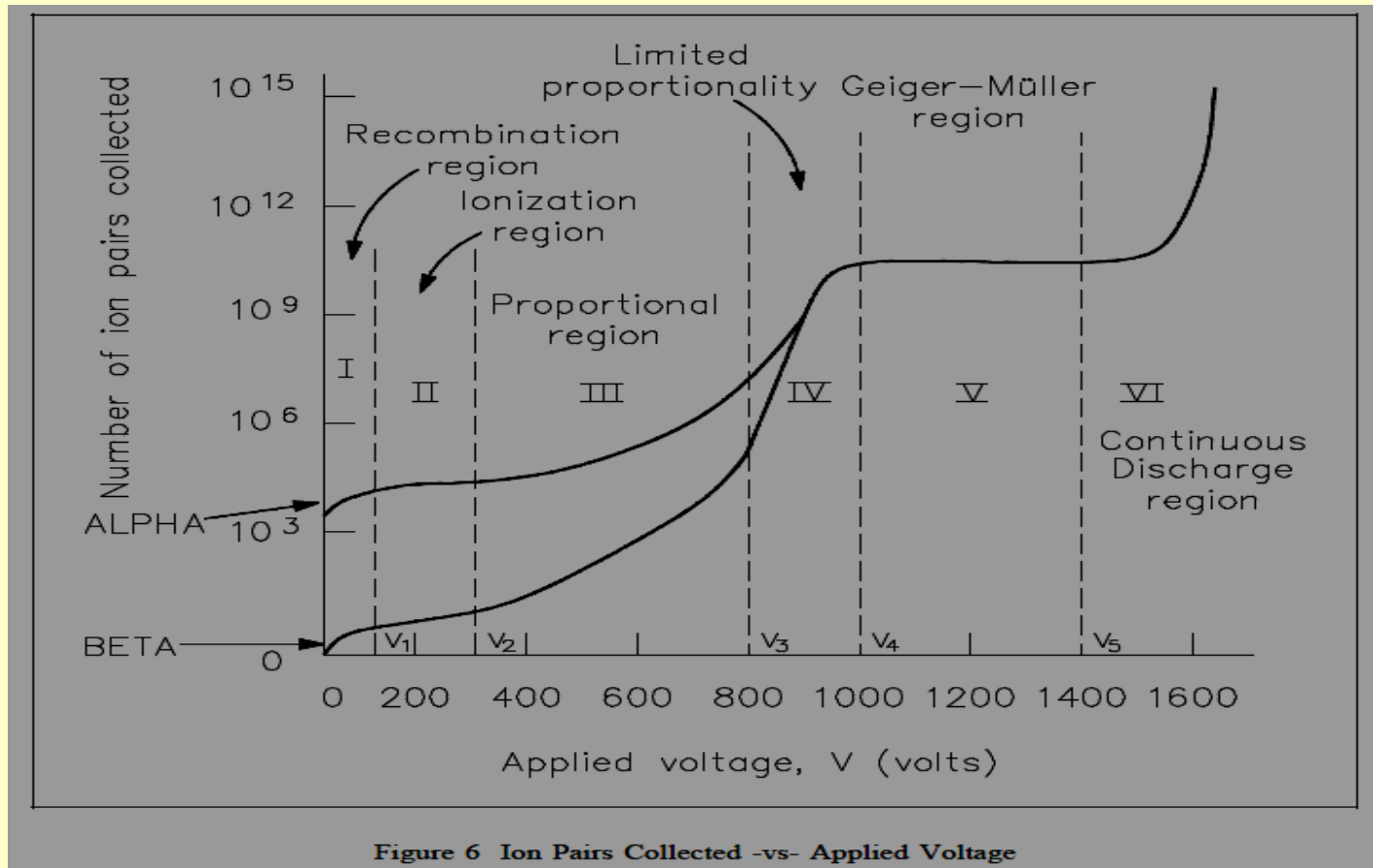


➡ voltage so low that **recombination takes place** before any charge is collected at the electrodes

➡ **no counters** operate in this region



Ionization Region



➡ voltage sufficient to ensure that **all electron-ion pairs** produced by the incident radiation are **collected**

➡ **no gas amplification** takes place



The Height of the Pulse

$$\Delta V = Ane/C$$

ΔV = pulse height (volts)

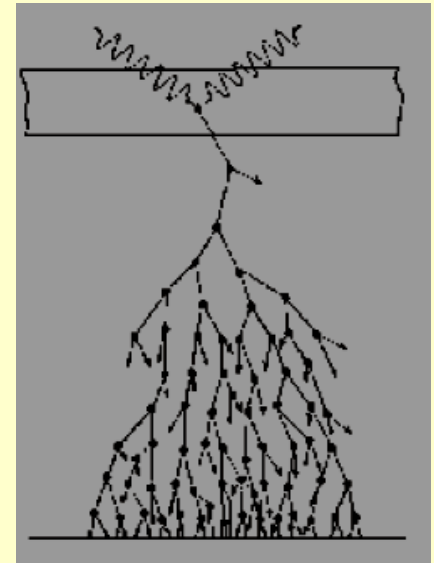
A = gas amplification factor

n = initial ionizing events

e = charge of the electron (1.602×10^{-19} coulombs)

C = detector capacitance (farads)

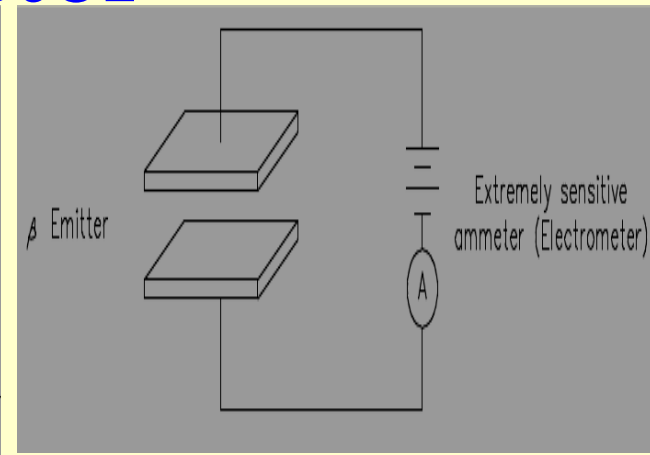
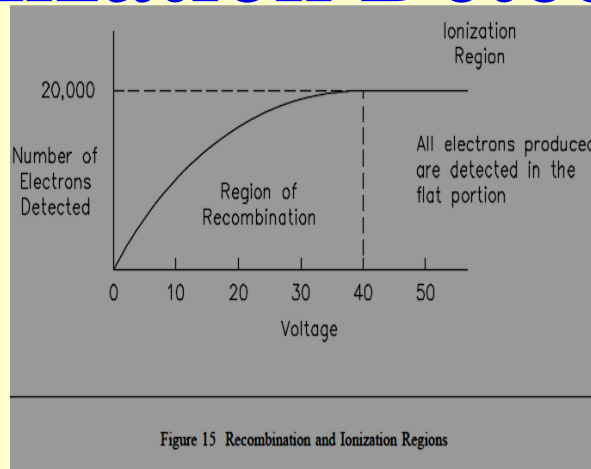
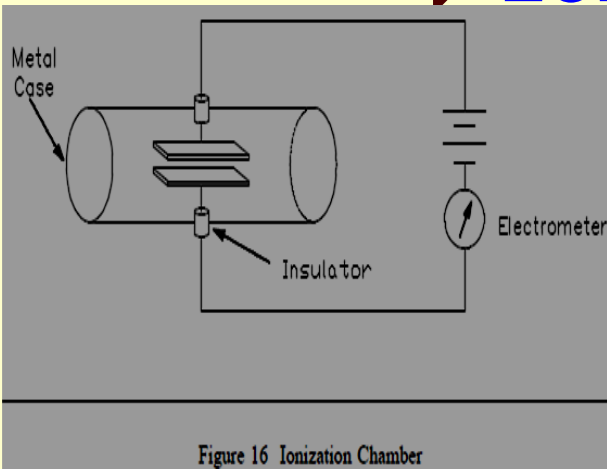
A = number of ions **collected** / number of **initial** ions



➤ **Fill Gas** - A noble gas which occupies 90 to 95% of the active volume of the detector. The noble gas is typically **helium, neon, or argon**.

➤ **Quench Gas** - A gas occupying 5 to 10% of the detector volume. The quench gas functions to **prevent the formation of spurious pulses**.

Ionization Detector



- ➡ When radiation enters an ionization chamber, the detector gas at the point of incident radiation **becomes ionized**.
- ➡ Some of the electrons have sufficient energy to **cause additional ionizations**.
- ➡ The electrons are **attracted** to the electrode by the voltage potential set up on the detector.
- ➡ If the voltage is set high enough, all of the electrons will reach the electrode **before recombination** takes place.

➤ Ionization Chamber Detector for Neutrons

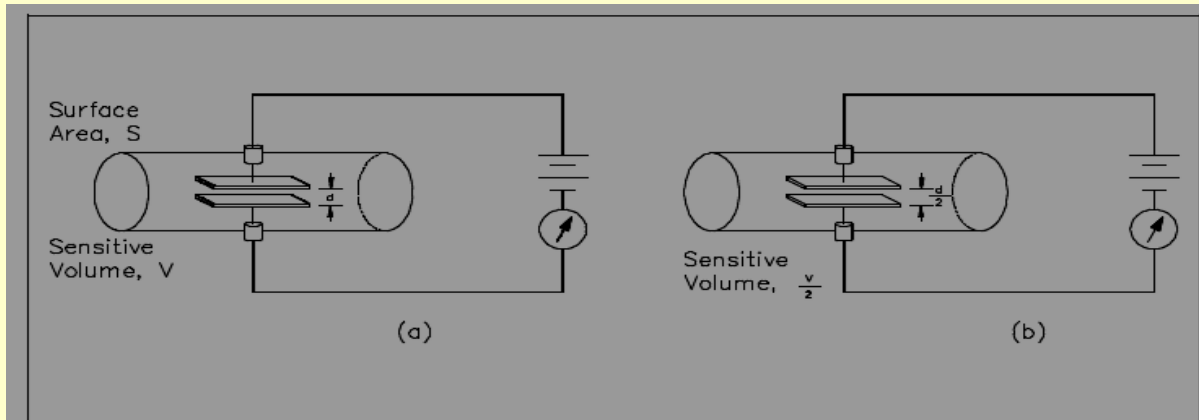
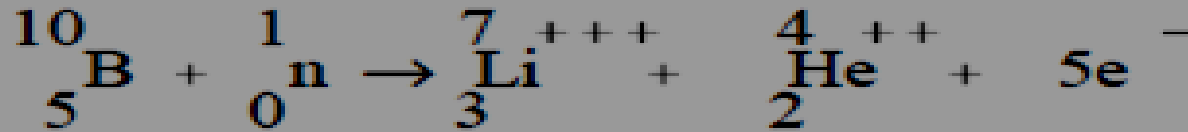


Figure 17 Minimizing Gamma Influence by Size and Volume

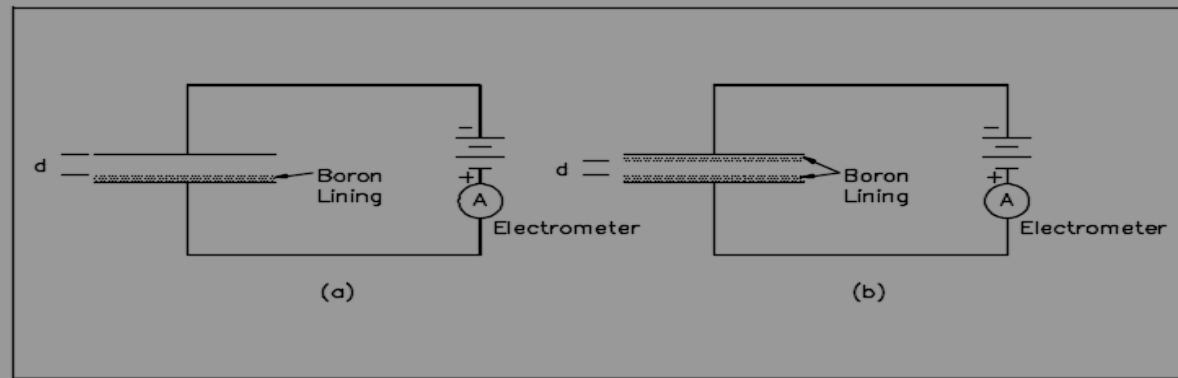
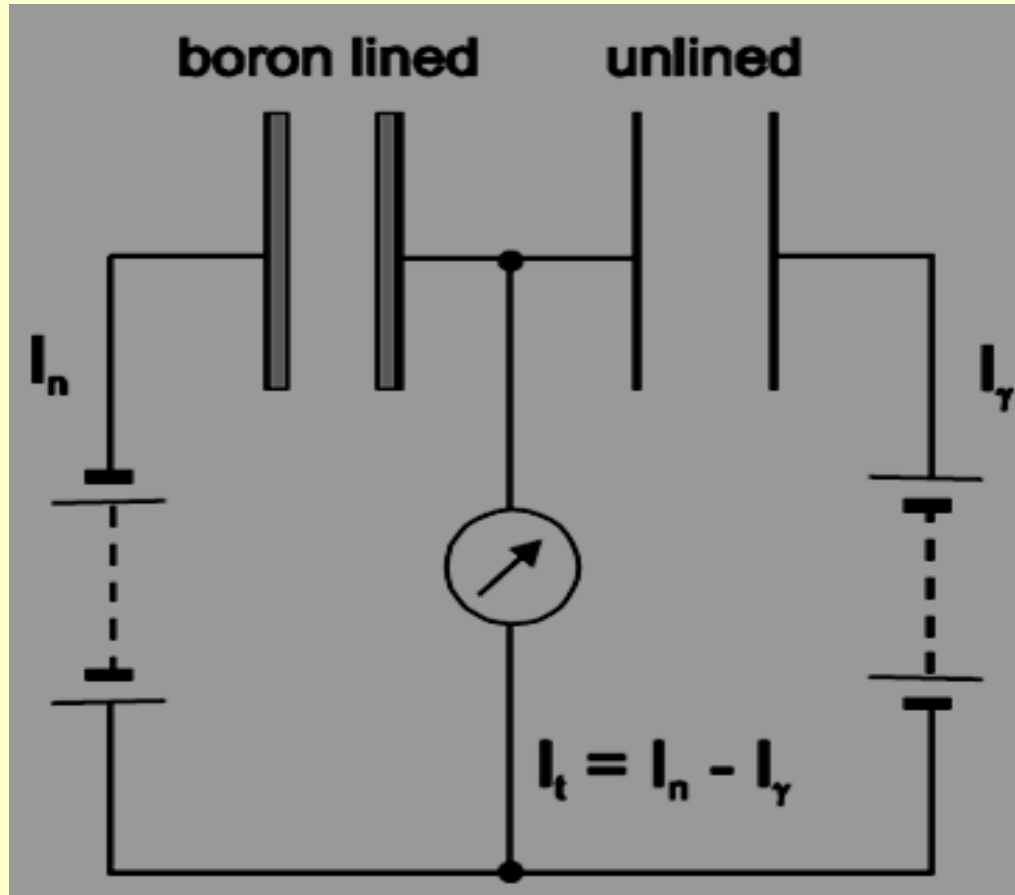


Figure 18 Minimizing Gamma Influence with Boron Coating Area

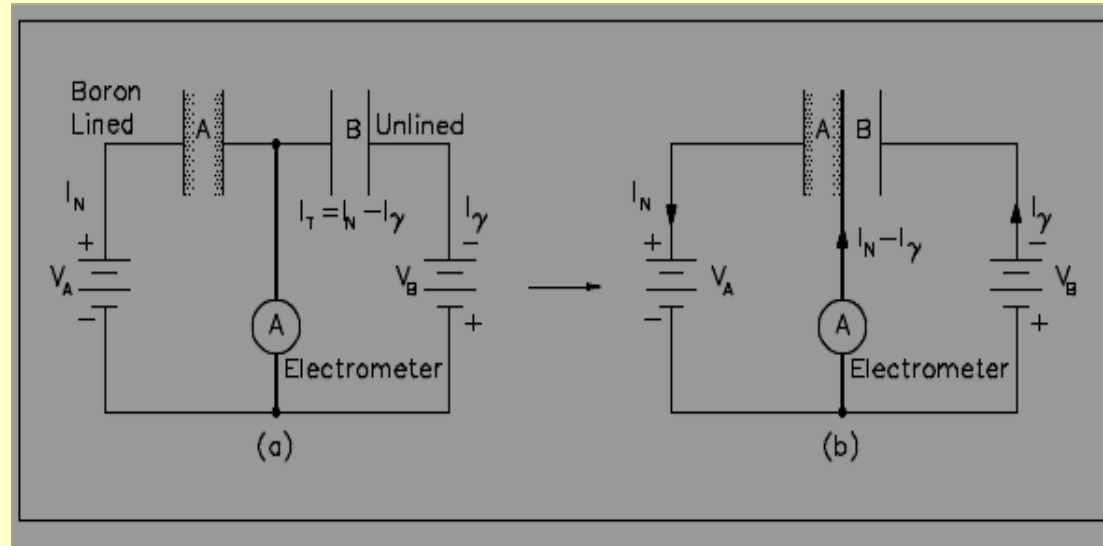
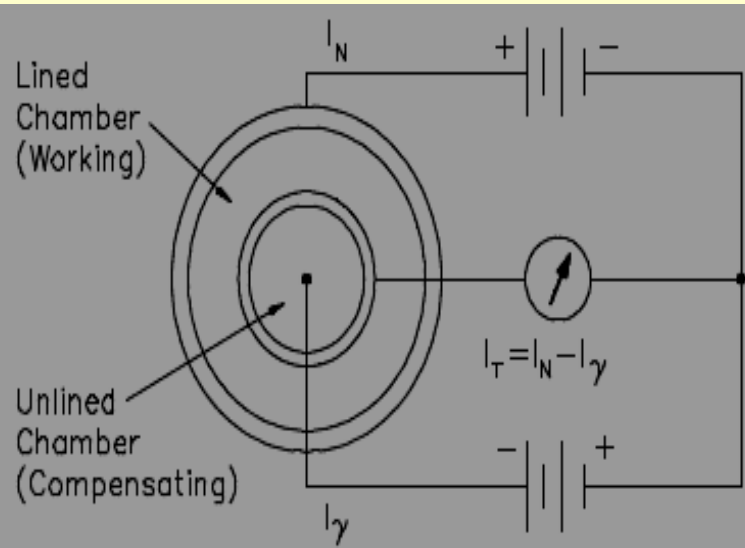
➤ Gamma sensitivity reduction is accomplished by either **reducing the amount of chamber gas or increasing the boron coated surface area.**

↪ Compensated Ionization Chamber



↪ The boron coated chamber is referred to as the **working chamber**; the uncoated chamber is called the **compensating chamber**.

➡ Compensated Ionization Chamber



- ❑ A compensated ion chamber has **two concentric cylinders**: a **boron-coated chamber** and an **uncoated chamber**.
- ❑ Both **gammas** and **neutrons** interact in the **boron-coated chamber**.
- ❑ Only **gammas** interact in the **uncoated chamber**.
- ❑ The **voltages to each chamber** are set so that the **current** from the **gammas** in the **boron-coated chamber** **cancels** the **current** from the **gammas** in the **uncoated chamber**.

↪ Compensated Ionization Chamber

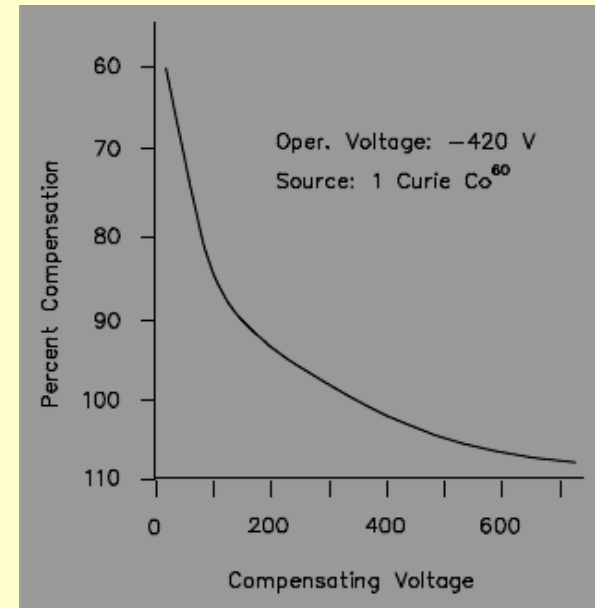
Percent compensation of a compensated ion chamber **gives the percentage of the gamma rays which are canceled out**. Percent compensation may be **calculated based on measured current, when the detector is exposed to gamma rays only**

$$\text{Percent Compensation} = 1 - \frac{I_{\text{measured}}}{I_{\text{operating}}} \times 100\%$$

where

I_{measured} = measured current (milliamps)

$I_{\text{operating}}$ = measured current with compensating voltage OFF (milliamps)



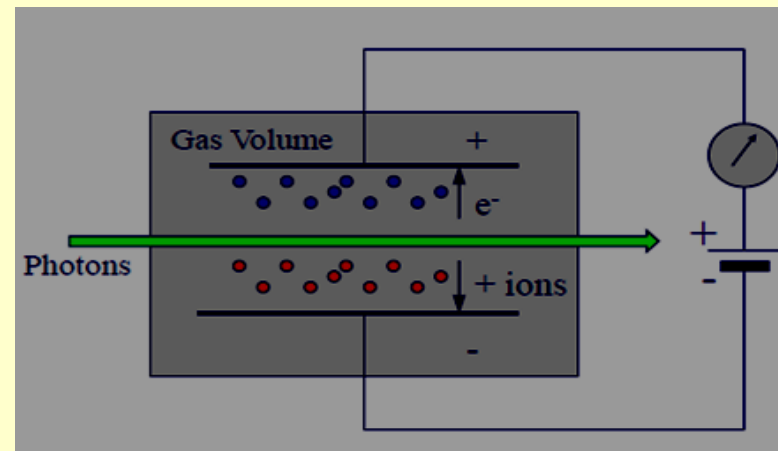
If measured current **is zero**, then **percent compensation is 100%**.

If measured current **is positive**, the percent compensation **is less than 100%**, and the chamber is **undercompensated**.

If the measured current **is negative**, the percent compensation **is greater than 100%**, and the chamber is **overcompensated**.

Ionization Detector Properties

- ❑ **output signal** proportional to the particle energy => identification of the type of particle and its energy
- ❑ **signal is not large** => only strongly ionizing particles (α , p, fission fragments, heavy ions) are detected
- ❑ **energy range** > 10 keV
- ❑ **pulse** counting ionization chamber and **integrating** ionization chamber
- ❑ **parallel plate or concentric** cylinder design, gas: air



- ❑ “**window**” for α and/or β
- ❑ detection of **neutrons**
- ❑ **gamma sensitivity is reduced** by **decreasing** the amount of chamber gas or **increasing** the boron coated surface area



Proportional Region

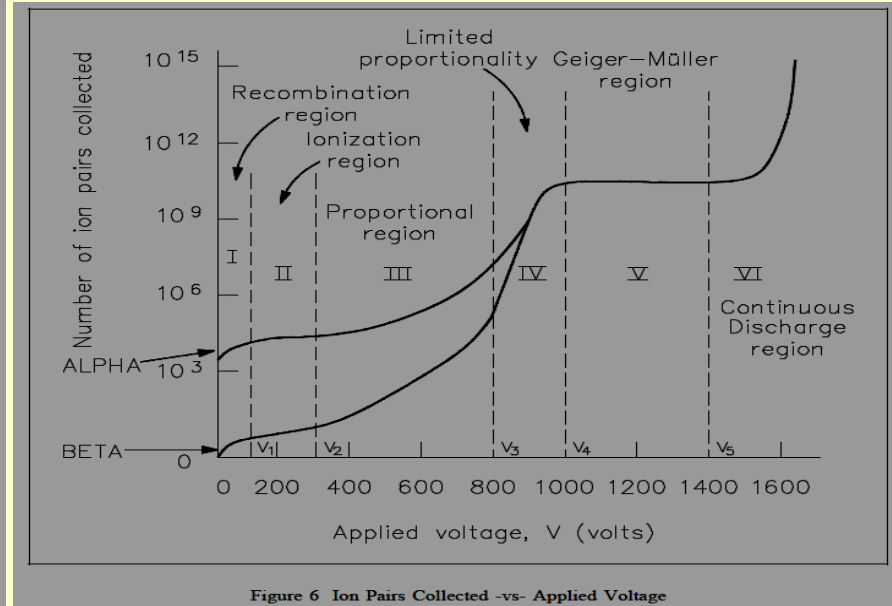
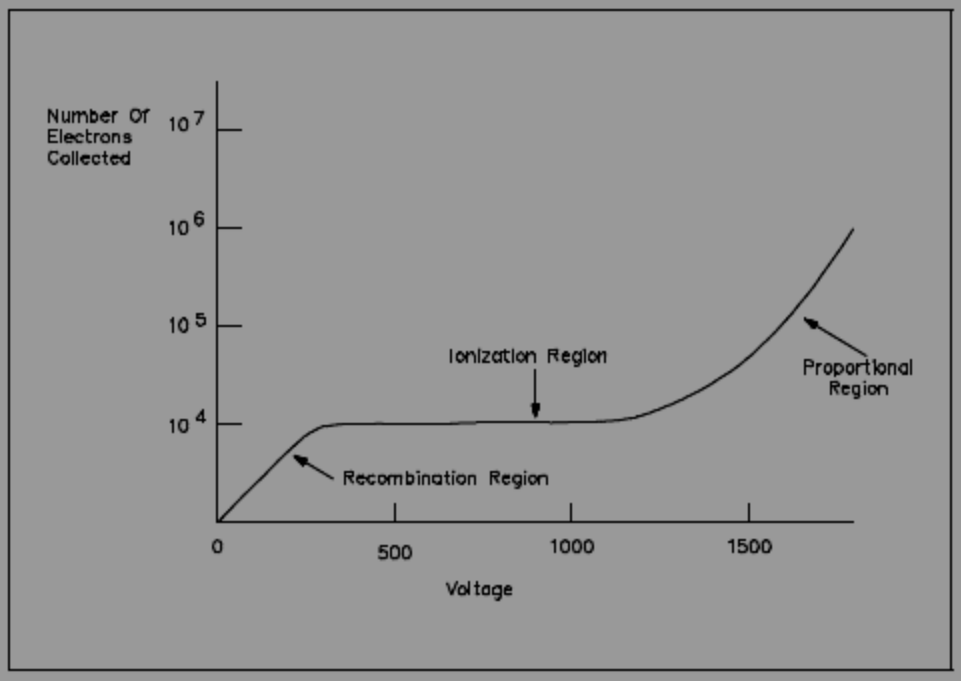


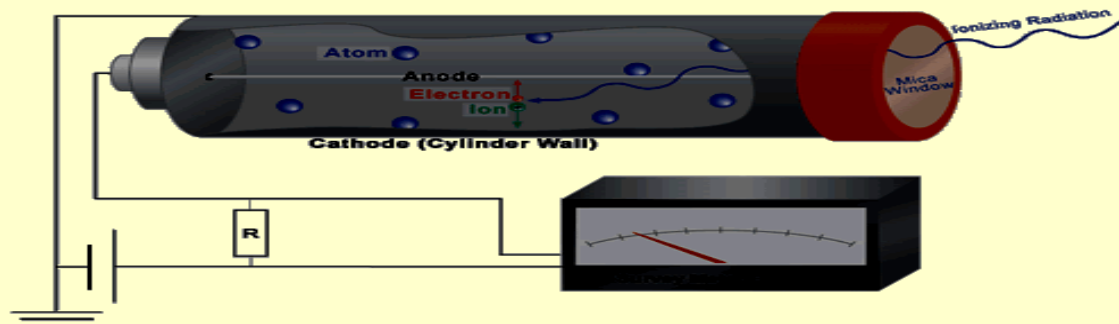
Figure 6 Ion Pairs Collected -vs- Applied Voltage

➡ **all** electron-ion pairs are collected

➡ gas amplification ($A \sim 10^3 - 10^4$) is **proportional** to the applied voltage

➤ Proportional Detector

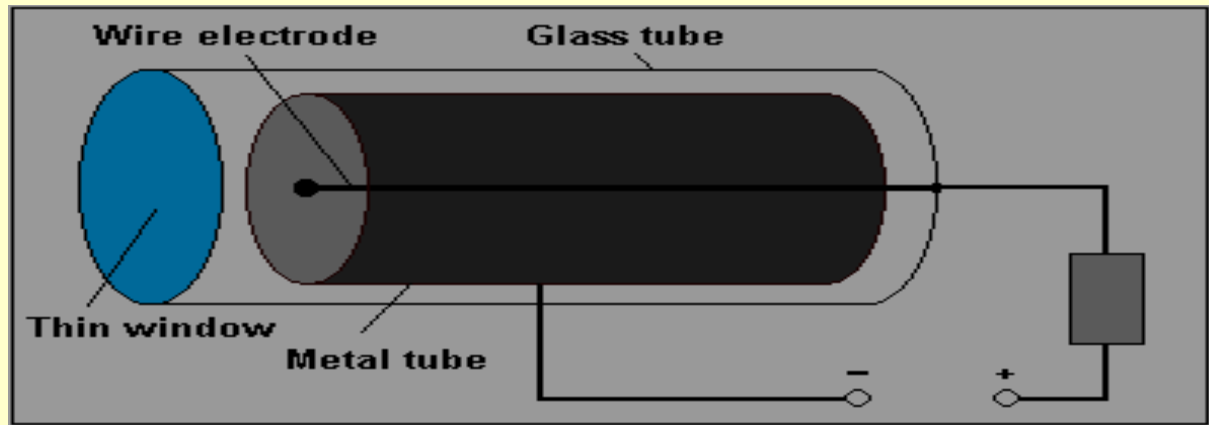
- When radiation enters a proportional counter, the detector gas, at the point of incident radiation, **becomes ionized**.
- The detector voltage is set so that the electrons cause secondary ionizations as they **accelerate** toward the electrode.
- The electrons produced from the secondary ionizations cause **additional** ionizations.



- This multiplication of electrons is called **gas amplification**.
- **Varying** the **detector voltage** within the proportional region increases or decreases the **gas amplification factor**.
- A quenching gas is added to give up electrons to the chamber gas so that **inaccuracies** are NOT introduced due to ionizations caused by the positive ion.

↪ Proportional Detector Properties

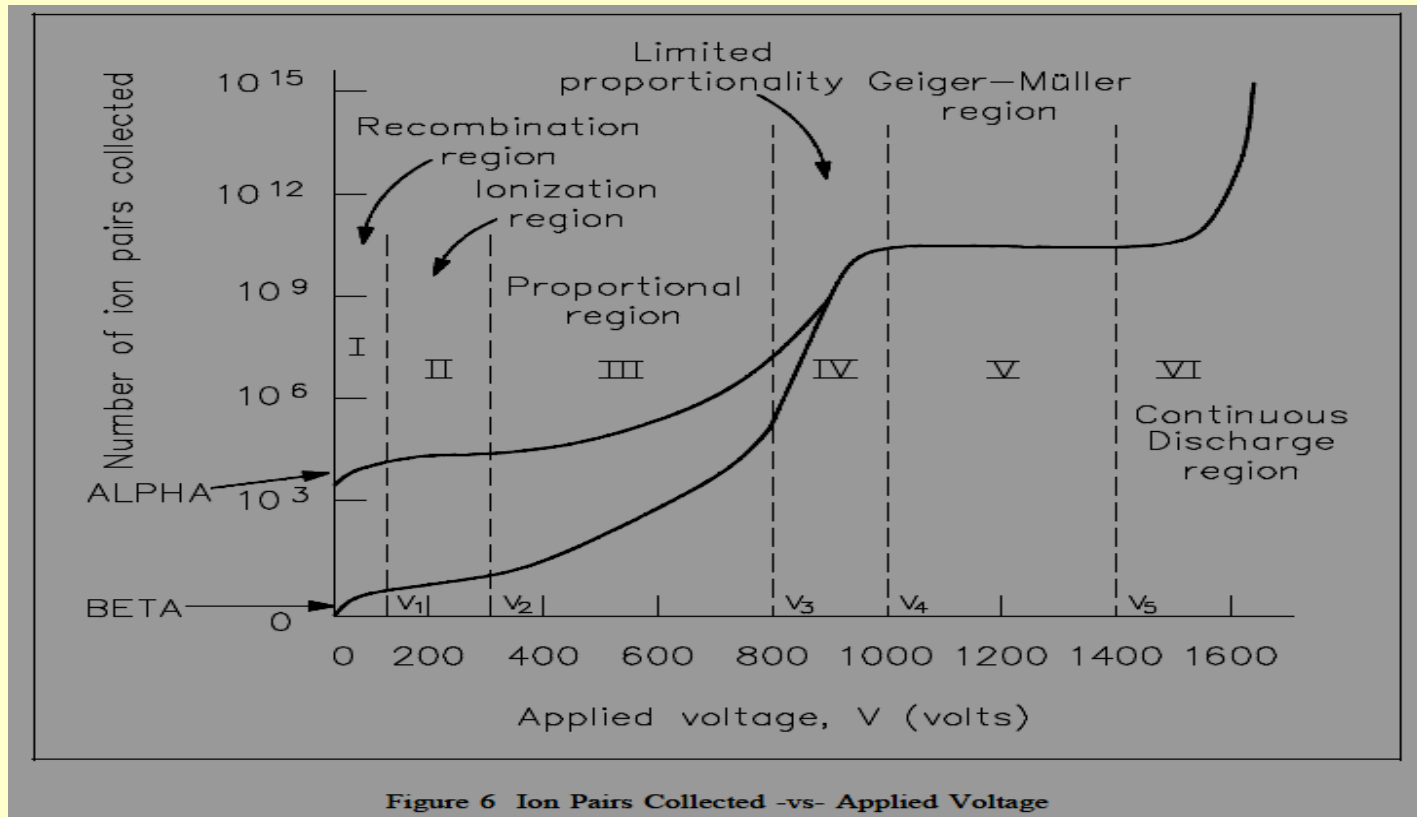
- ▣ **identification** of the any type of particle (α , β , γ) and its **energy**
- ▣ **gas**: **helium, argon**, (α , β , γ) or **boron trifluoride** (neutrons)
- ▣ **free electrons** are produced by **secondary ionization, photoelectric interactions, and bombardment of the cathode surface by positive ions**



- ▣ **detection of low energy particles** (< 10 keV) is possible due to **internal amplification**
- ▣ **quenching gas** (methan) \Rightarrow limited lifetime \Rightarrow **gas flow counters**



Limited Proportional Region



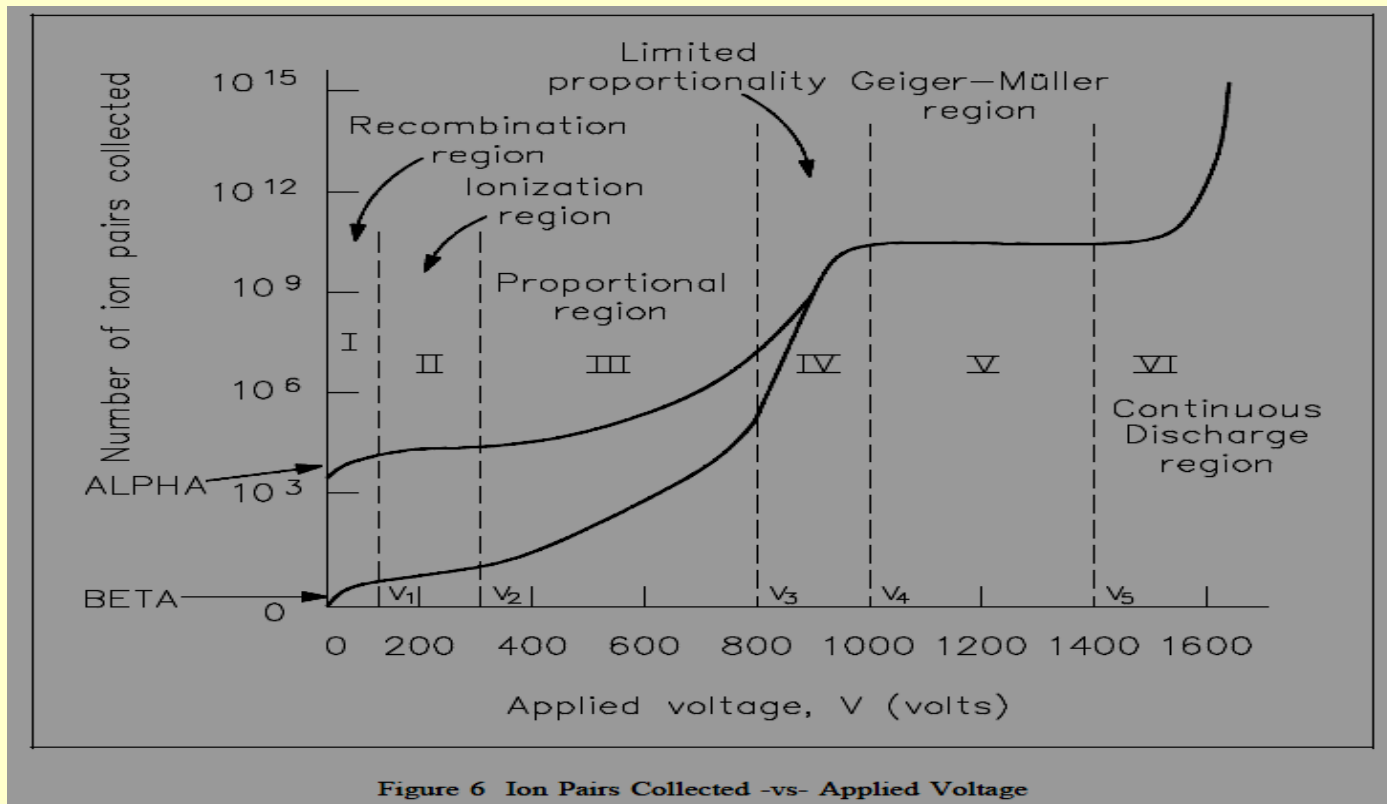
➡ additional processes occur leading to increased ionization

➡ positive ions remain near their point of origin, further avalanches are impossible

➡ no counters operate in this region



Geiger-Muller Region



➡ electron-ion pair production is **independent** on the type of incident radiation

➡ electric field strength so high that the discharge continues to spread until **amplification cannot occur**, due to a dense positive ion sheath surrounding the anode

➤ Geiger-Muller Detector

➤ The voltage of a Geiger-Muller (G-M) detector is set so that **any incident radiation produces the same number of electrons**.

➤ As long as voltage remains in the G-M region, **electron production is independent of operating voltage** and the **initial number of electrons** produced by the **incident radiation**.

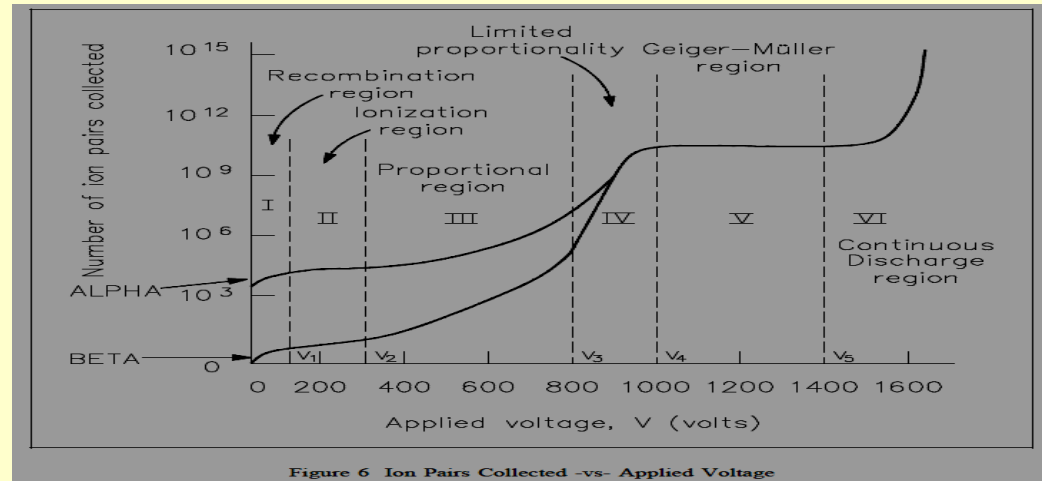
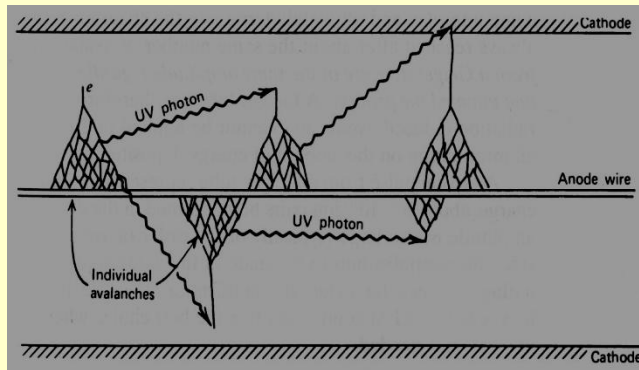


Figure 6 Ion Pairs Collected -vs- Applied Voltage

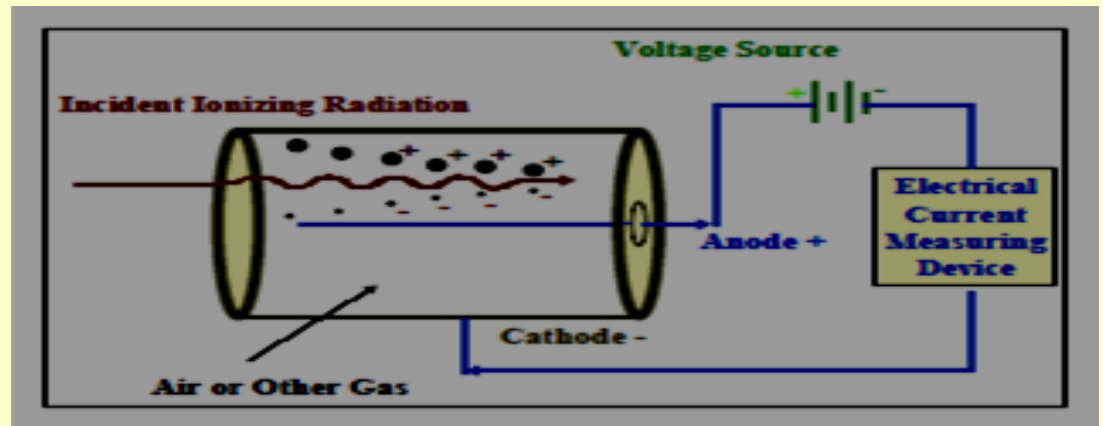
➤ The **operation voltage** causes a **large number of ionizations** to occur near the central electrode as the **electrons approach**.

➤ The large number of positive ions form a **positive ion sheath** which prevents additional electrons from reaching the electrode.

➤ A **quenching gas** is used in order to prevent a secondary pulse due to ionization by the positive ions.

↪ Geiger-Muller Properties

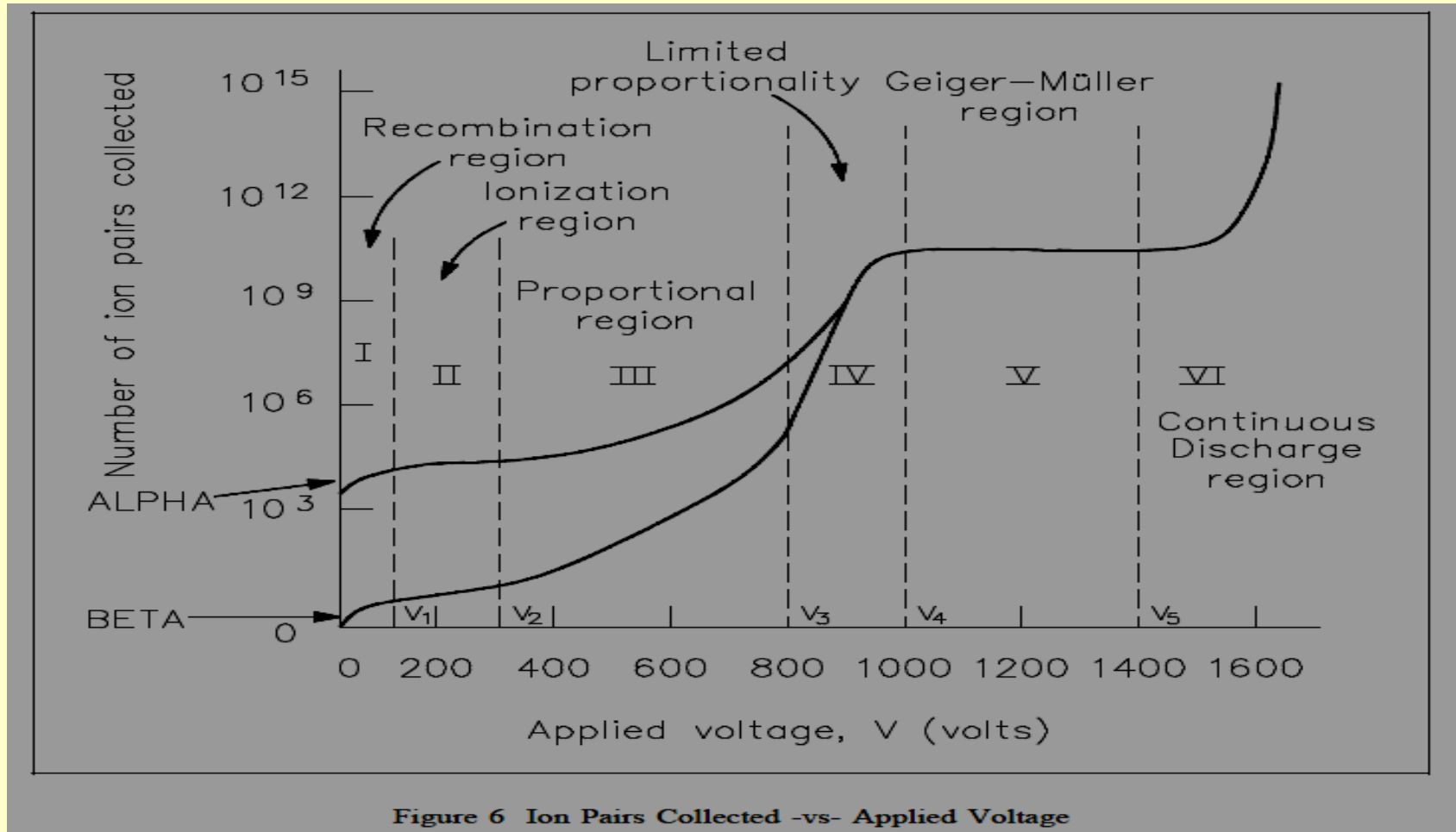
- ❑ **identification** of the type of radiation is impossible
- ❑ **information** only about the number of particles
- ❑ **positive ion sheath around the central wire** => reduced field strength -> quenching gas



- ❑ **finite lifetime** due to degradation of the filled-in gas
- ❑ use limited to **low counting rates** due to large dead time
- ❑ **sensitive detectors** -> portable instrumentation, simple counting circuit, ability to detect low level radiation



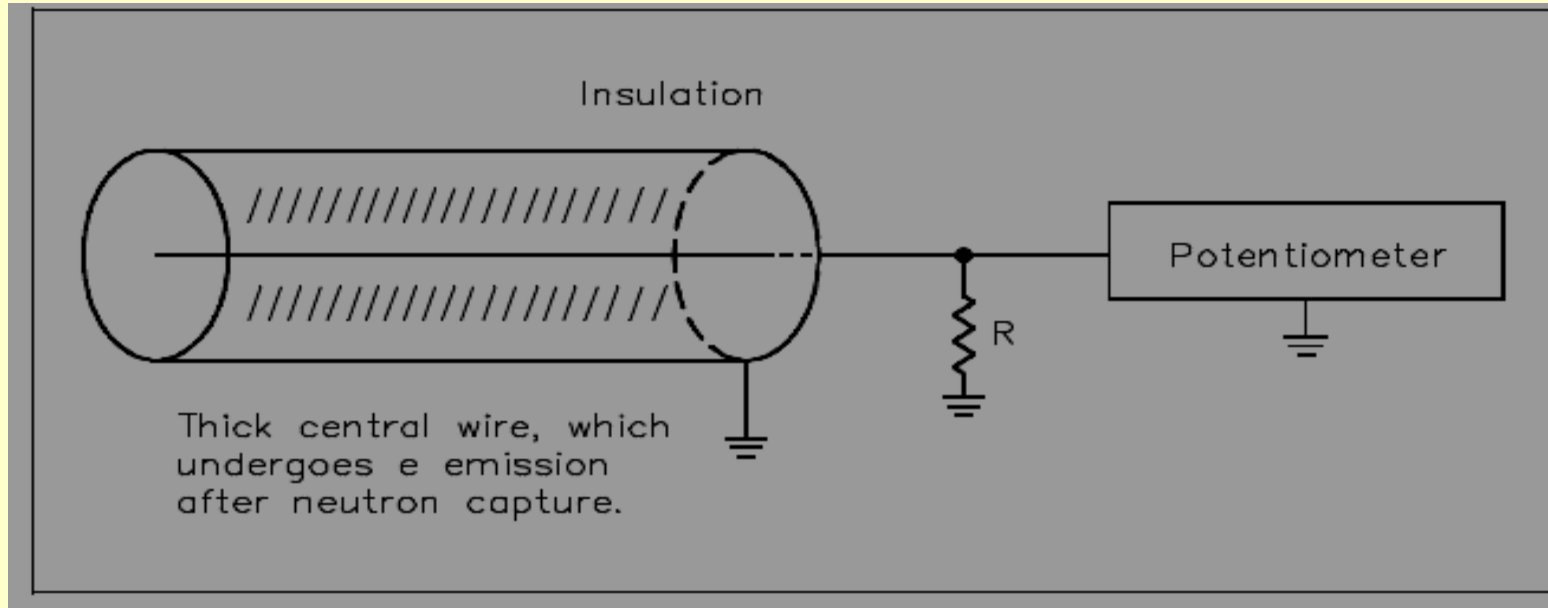
Continuous Discharge Region



➡ the applied voltage is so **high** that, **once ionization takes place**, there is a **continuous discharge**

➤ Miscellaneous Detectors

1- Self-Powered Neutron Detector



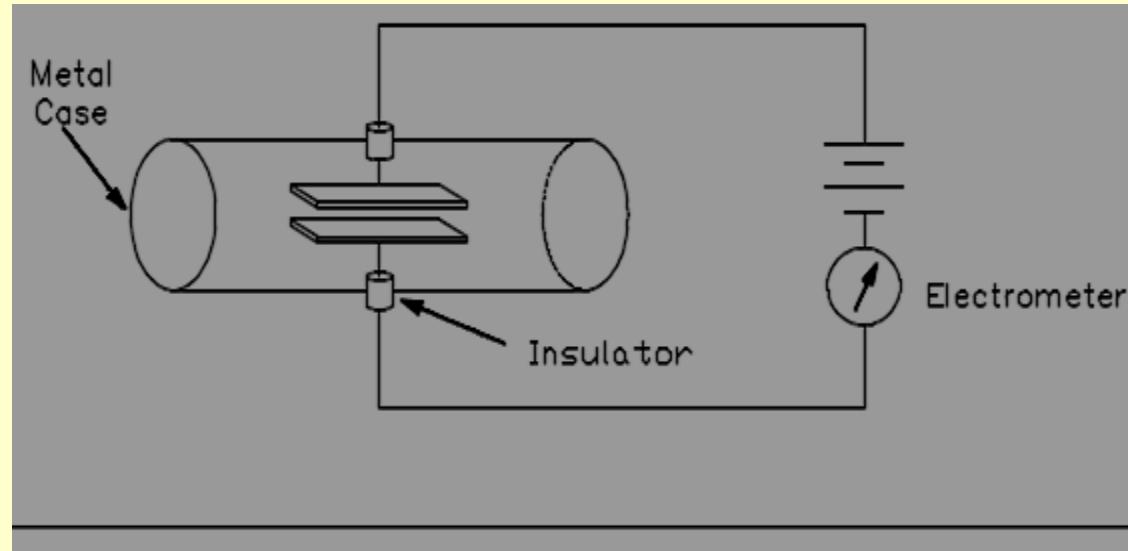
➤ The **central wire**, made of a **neutron-absorbing** material, absorbs a neutron and undergoes beta decay.

➤ As more **beta decays occur**, the remaining atoms cause the wire to become more **positively charged**.

➤ The voltage potential set up causes a **current flow** in a resistor, which is measured by either a millivoltmeter or electrometer.

↪ Miscellaneous Detectors

2- Wide Range Fission Chamber



↪ Neutrons interact with the **U^{235} coated chamber** causing fission of the **U^{235}** .

↪ A highly positive charged **fission fragment interacts** with the detector gas and **causes ionizations**.

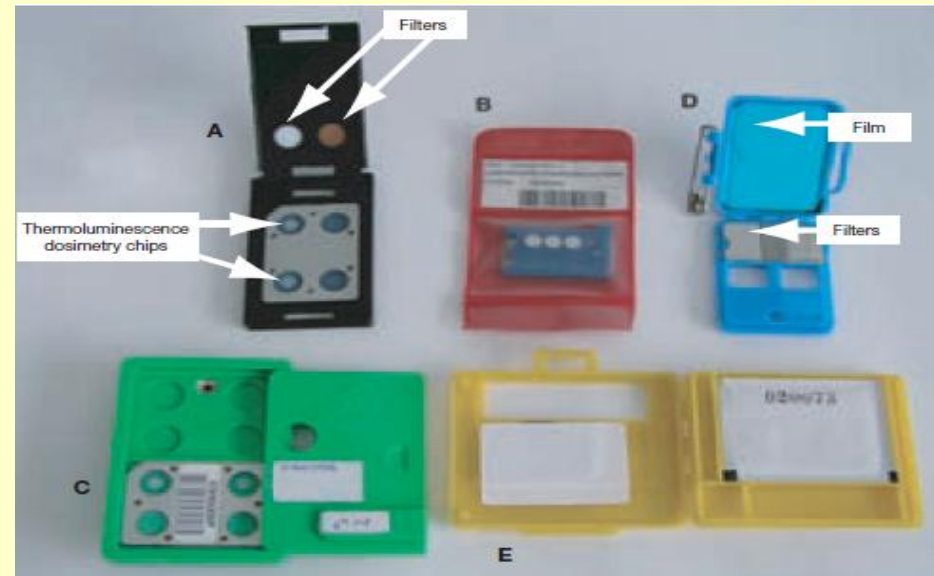
↪ The **electrons produced** are collected as pulses on the electrode.

➡ Miscellaneous Detectors

3- Activation Foils and Flux Wires

➡ The **wire** is inserted directly into the core and becomes **activated** by the **neutron flux**.

➡ When the **is reached**, the **wire is removed** from the core and **co** desired activation time counted.



4- Photographic Film

➡ Detects total radiation dose by **darkening**; **film darkness determines** overall exposure.

➡ Fast neutron exposure determined by **counting individual proton recoil tracks**.